

# Decyl laurate

<b>Other names:</b>	Decyl dodecanoate
<b>Inchi:</b>	InChI=1S/C22H44O2/c1-3-5-7-9-11-13-14-16-18-20-22(23)24-21-19-17-15-12-10-8-6-4-2
<b>InchiKey:</b>	RAPXDXJBAYUBHI-UHFFFAOYSA-N
<b>Formula:</b>	C22H44O2
<b>SMILES:</b>	CCCCCCCCCCCC(=O)CCCCCCCCCCC
<b>Mol. weight [g/mol]:</b>	340.58
<b>CAS:</b>	36528-28-6

## Physical Properties

Property code	Value	Unit	Source
gf	-99.56	kJ/mol	Joback Method
hf	-742.21	kJ/mol	Joback Method
hfus	55.52	kJ/mol	Joback Method
hvap	73.72	kJ/mol	Joback Method
log10ws	-7.89		Crippen Method
logp	7.591		Crippen Method
mcvol	328.280	ml/mol	McGowan Method
pc	927.24	kPa	Joback Method
rinpol	2356.00		NIST Webbook
rinpol	2371.70		NIST Webbook
tb	779.05	K	Joback Method
tc	955.69	K	Joback Method
tf	409.86	K	Joback Method
vc	1.292	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1031.01	J/molxK	779.05	Joback Method
cpg	1051.60	J/molxK	808.49	Joback Method
cpg	1071.17	J/molxK	837.93	Joback Method
cpg	1089.74	J/molxK	867.37	Joback Method
cpg	1107.35	J/molxK	896.81	Joback Method
cpg	1124.02	J/molxK	926.25	Joback Method

cpg	1139.78	J/molxK	955.69	Joback Method
dvisc	0.0013386	Paxs	409.86	Joback Method
dvisc	0.0005575	Paxs	471.39	Joback Method
dvisc	0.0002843	Paxs	532.92	Joback Method
dvisc	0.0001666	Paxs	594.45	Joback Method
dvisc	0.0001080	Paxs	655.99	Joback Method
dvisc	0.0000754	Paxs	717.52	Joback Method
dvisc	0.0000557	Paxs	779.05	Joback Method
hfust	63.67	kJ/mol	293.20	NIST Webbook

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C36528286&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C36528286&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hfust:</b>	Enthalpy of fusion at a given temperature
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mccvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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