

# 4-Tridecanone

<b>Other names:</b>	n-Nonyl n-propyl ketone tridecan-4-one
<b>Inchi:</b>	InChI=1S/C13H26O/c1-3-5-6-7-8-9-10-12-13(14)11-4-2/h3-12H2,1-2H3
<b>InchiKey:</b>	JWSRUPAFLPWLNO-UHFFFAOYSA-N
<b>Formula:</b>	C13H26O
<b>SMILES:</b>	CCCCCCCCC(=O)CCC
<b>Mol. weight [g/mol]:</b>	198.34
<b>CAS:</b>	26215-90-7

## Physical Properties

Property code	Value	Unit	Source
gf	-70.34	kJ/mol	Joback Method
hf	-424.23	kJ/mol	Joback Method
hfus	31.02	kJ/mol	Joback Method
hvap	51.28	kJ/mol	Joback Method
log10ws	-4.54		Crippen Method
logp	4.496		Crippen Method
mvol	195.600	ml/mol	McGowan Method
pc	1731.78	kPa	Joback Method
rhoc	241.98 ± 7.93	kg/m <sup>3</sup>	NIST Webbook
tb	550.71	K	Joback Method
tc	711.60 ± 3.50	K	NIST Webbook
tf	286.20	K	Joback Method
vc	0.769	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	486.12	J/mol×K	550.71	Joback Method
cpg	562.38	J/mol×K	691.58	Joback Method
cpg	548.43	J/mol×K	663.40	Joback Method
cpg	533.84	J/mol×K	635.23	Joback Method
cpg	518.60	J/mol×K	607.06	Joback Method
cpg	502.70	J/mol×K	578.88	Joback Method

cpg	575.73	J/molxK	719.75	Joback Method
dvisc	0.0002125	Paxs	550.71	Joback Method
dvisc	0.0002816	Paxs	506.62	Joback Method
dvisc	0.0003936	Paxs	462.54	Joback Method
dvisc	0.0005905	Paxs	418.46	Joback Method
dvisc	0.0009747	Paxs	374.37	Joback Method
dvisc	0.0018390	Paxs	330.28	Joback Method
dvisc	0.0042194	Paxs	286.20	Joback Method

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.52310e+01
Coeff. B	-4.71493e+03
Coeff. C	-8.95600e+01
Temperature range (K), min.	405.08
Temperature range (K), max.	564.88

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C26215907&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C26215907&amp;Units=SI</a>
<b>The Yaws Handbook of Vapor Pressure: Crippen Method:</b>	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions

<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>rhoc:</b>	Critical density
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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