

# Glutaric acid, ditetrahydrofurfuryl ester

**Inchi:** InChI=1S/C15H24O6/c16-14(20-10-12-4-2-8-18-12)6-1-7-15(17)21-11-13-5-3-9-19-13/h1-15  
**InchiKey:** VLQQQWCPHFOKKR-UHFFFAOYSA-N  
**Formula:** C15H24O6  
**SMILES:** O=C(CCCC(=O)OCC1CCCO1)OCC1CCCO1  
**Mol. weight [g/mol]:** 300.35

## Physical Properties

Property code	Value	Unit	Source
gf	-491.56	kJ/mol	Joback Method
hf	-985.57	kJ/mol	Joback Method
hfus	44.01	kJ/mol	Joback Method
hvap	76.83	kJ/mol	Joback Method
log10ws	-2.01		Crippen Method
logp	1.601		Crippen Method
mcvol	227.110	ml/mol	McGowan Method
pc	2021.76	kPa	Joback Method
rinpol	2369.00		NIST Webbook
rinpol	2369.00		NIST Webbook
tb	779.64	K	Joback Method
tc	991.42	K	Joback Method
tf	478.07	K	Joback Method
vc	0.848	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	727.20	J/molxK	779.64	Joback Method
cpg	744.33	J/molxK	814.94	Joback Method
cpg	760.18	J/molxK	850.23	Joback Method
cpg	774.78	J/molxK	885.53	Joback Method
cpg	788.16	J/molxK	920.82	Joback Method
cpg	800.35	J/molxK	956.12	Joback Method
cpg	811.36	J/molxK	991.42	Joback Method
dvisc	0.0016040	Paxs	478.07	Joback Method

dvisc	0.0009308	Paxs	528.33	Joback Method
dvisc	0.0005937	Paxs	578.59	Joback Method
dvisc	0.0004069	Paxs	628.86	Joback Method
dvisc	0.0002949	Paxs	679.12	Joback Method
dvisc	0.0002234	Paxs	729.38	Joback Method
dvisc	0.0001755	Paxs	779.64	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U359674&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U359674&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>m<sub>cvol</sub>:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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