

# Caffeic acid

<b>Other names:</b>	2-Propenoic acid, 3-(3,4-dihydroxyphenyl)- Cinnamic acid, 3,4-dihydroxy- 3-(3,4-Dihydroxyphenyl)-2-propenoic acid 3-(3,4-Dihydroxyphenyl)propenoic acid 3,4-Dihydroxybenzeneacrylic acid 3,4-Dihydroxycinnamic acid 4-(2-Carboxyethenyl)-1,2-dihydroxybenzene 4-(2'-Carboxyvinyl)-1,2-dihydroxybenzene NSC 57197 3,4-Dihydroxycinnamic acid (caffeic acid)
<b>Inchi:</b>	InChI=1S/C9H8O4/c10-7-3-1-6(5-8(7)11)2-4-9(12)13/h1-5,10-11H,(H,12,13)/b4-2+
<b>InchiKey:</b>	QAIPRVGONGVQAS-DUXPYHPUSA-N
<b>Formula:</b>	C9H8O4
<b>SMILES:</b>	O=C(O)C=Cc1ccc(O)c(O)c1
<b>Mol. weight [g/mol]:</b>	180.16
<b>CAS:</b>	331-39-5

## Physical Properties

Property code	Value	Unit	Source
gf	-357.45	kJ/mol	Joback Method
hf	-494.77	kJ/mol	Joback Method
hfus	30.56	kJ/mol	Joback Method
hvap	87.31	kJ/mol	Joback Method
log10ws	-0.91		Crippen Method
logp	1.196		Crippen Method
mcvol	128.790	ml/mol	McGowan Method
pc	6441.16	kPa	Joback Method
rinpol	1478.00		NIST Webbook
tb	743.45	K	Joback Method
tc	972.70	K	Joback Method
tf	546.72	K	Joback Method
vc	0.368	m <sup>3</sup> /kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	381.42	J/molxK	972.70	Joback Method
cpg	373.93	J/molxK	934.49	Joback Method
cpg	366.78	J/molxK	896.28	Joback Method
cpg	359.81	J/molxK	858.07	Joback Method
cpg	352.90	J/molxK	819.87	Joback Method
cpg	345.89	J/molxK	781.66	Joback Method
cpg	338.64	J/molxK	743.45	Joback Method
dvisc	0.0000171	Paxs	546.72	Joback Method
dvisc	0.0000003	Paxs	743.45	Joback Method
dvisc	0.0000005	Paxs	710.66	Joback Method
dvisc	0.0000010	Paxs	677.87	Joback Method
dvisc	0.0000018	Paxs	645.09	Joback Method
dvisc	0.0000035	Paxs	612.30	Joback Method
dvisc	0.0000074	Paxs	579.51	Joback Method
hsubt	170.20 ± 4.60	kJ/mol	416.50	NIST Webbook

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C331395&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C331395&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hsubt:</b>	Enthalpy of sublimation at a given temperature
<b>hvap:</b>	Enthalpy of vaporization at standard conditions

<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpola:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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