

# Pentanoic acid, 5-bromo-, ethyl ester

<b>Other names:</b>	Ethyl 5-bromopentanoate Ethyl 5-bromovalerate Valeric acid, 5-bromo-, ethyl ester Ethyl «omega»-bromovalerate 5-Bromovaleric acid, ethyl ester
<b>Inchi:</b>	InChI=1S/C7H13BrO2/c1-2-10-7(9)5-3-4-6-8/h2-6H2,1H3
<b>InchiKey:</b>	AFRWBGJRWRHQOV-UHFFFAOYSA-N
<b>Formula:</b>	C7H13BrO2
<b>SMILES:</b>	CCOC(=O)CCCCBr
<b>Mol. weight [g/mol]:</b>	209.08
<b>CAS:</b>	14660-52-7

## Physical Properties

Property code	Value	Unit	Source
gf	-211.54	kJ/mol	Joback Method
hf	-406.28	kJ/mol	Joback Method
hfus	21.96	kJ/mol	Joback Method
hvap	46.77	kJ/mol	Joback Method
log10ws	-2.05		Crippen Method
logp	2.115		Crippen Method
mvol	134.430	ml/mol	McGowan Method
pc	3173.97	kPa	Joback Method
rinpol	1240.00		NIST Webbook
tb	502.01	K	Joback Method
tc	694.06	K	Joback Method
tf	300.61	K	Joback Method
vc	0.513	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	276.81	J/mol×K	502.01	Joback Method
cpg	326.65	J/mol×K	662.05	Joback Method
cpg	317.59	J/mol×K	630.04	Joback Method

cpg	308.09	J/molxK	598.04	Joback Method
cpg	298.13	J/molxK	566.03	Joback Method
cpg	287.70	J/molxK	534.02	Joback Method
cpg	335.26	J/molxK	694.06	Joback Method
dvisc	0.0002990	Paxs	502.01	Joback Method
dvisc	0.0003769	Paxs	468.44	Joback Method
dvisc	0.0004925	Paxs	434.88	Joback Method
dvisc	0.0006729	Paxs	401.31	Joback Method
dvisc	0.0009732	Paxs	367.74	Joback Method
dvisc	0.0015160	Paxs	334.18	Joback Method
dvisc	0.0026071	Paxs	300.61	Joback Method

## Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	379.70	K	1.60	NIST Webbook

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C14660527&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C14660527&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient

<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.cheméo.com/cid/44-875-2/Pentanoic-acid-5-bromo-ethyl-ester.pdf>

Generated by Cheméo on 2024-04-20 03:50:04.125754568 +0000 UTC m=+15874253.046331884.

Cheméo (<https://www.cheméo.com>) is the biggest free database of chemical and physical data for the process industry.