

D-Alanine, N-neopentylloxycarbonyl-, hexyl ester

Inchi:	InChI=1S/C15H29NO4/c1-6-7-8-9-10-19-13(17)12(2)16-14(18)20-11-15(3,4)5/h12H,6-11
InchiKey:	NKUZRLRYPFTVJC-UHFFFAOYSA-N
Formula:	C15H29NO4
SMILES:	CCCCCOC(=O)C(C)NC(=O)OCC(C)(C)C
Mol. weight [g/mol]:	287.40

Physical Properties

Property code	Value	Unit	Source
gf	-302.63	kJ/mol	Joback Method
hf	-803.09	kJ/mol	Joback Method
hfus	34.34	kJ/mol	Joback Method
hvap	72.05	kJ/mol	Joback Method
log10ws	-3.86		Crippen Method
logp	3.271		Crippen Method
mcvol	247.070	ml/mol	McGowan Method
pc	1567.23	kPa	Joback Method
rinpol	1846.00		NIST Webbook
tb	741.68	K	Joback Method
tc	929.19	K	Joback Method
tf	443.21	K	Joback Method
vc	0.942	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	739.80	J/molxK	741.68	Joback Method
cpg	755.98	J/molxK	772.93	Joback Method
cpg	771.24	J/molxK	804.18	Joback Method
cpg	785.59	J/molxK	835.43	Joback Method
cpg	799.06	J/molxK	866.69	Joback Method
cpg	811.67	J/molxK	897.94	Joback Method
cpg	823.44	J/molxK	929.19	Joback Method

Sources

McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=U347764&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mccvol:	McGowan's characteristic volume
pc:	Critical Pressure
rinpol:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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