

# 4-Ethylbenzoic acid, pentyl ester

<b>Inchi:</b>	InChI=1S/C14H20O2/c1-3-5-6-11-16-14(15)13-9-7-12(4-2)8-10-13/h7-10H,3-6,11H2,1-2H
<b>InchiKey:</b>	ZQAAPECGIMFVID-UHFFFAOYSA-N
<b>Formula:</b>	C14H20O2
<b>SMILES:</b>	CCCCCOC(=O)c1ccc(CC)cc1
<b>Mol. weight [g/mol]:</b>	220.31

## Physical Properties

Property code	Value	Unit	Source
gf	-64.14	kJ/mol	Joback Method
hf	-352.03	kJ/mol	Joback Method
hfus	28.46	kJ/mol	Joback Method
hvap	58.85	kJ/mol	Joback Method
log10ws	-4.19		Crippen Method
logp	3.596		Crippen Method
mcvol	191.800	ml/mol	McGowan Method
pc	2066.12	kPa	Joback Method
rinpol	1717.80		NIST Webbook
rinpol	1717.80		NIST Webbook
tb	627.67	K	Joback Method
tc	828.65	K	Joback Method
tf	358.64	K	Joback Method
vc	0.736	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	493.85	J/molxK	627.67	Joback Method
cpg	509.98	J/molxK	661.17	Joback Method
cpg	525.23	J/molxK	694.66	Joback Method
cpg	539.64	J/molxK	728.16	Joback Method
cpg	553.21	J/molxK	761.65	Joback Method
cpg	565.96	J/molxK	795.15	Joback Method
cpg	577.93	J/molxK	828.65	Joback Method
dvisc	0.0015939	Paxs	358.64	Joback Method

dvisc	0.0008632	Paxs	403.48	Joback Method
dvisc	0.0005285	Paxs	448.32	Joback Method
dvisc	0.0003538	Paxs	493.16	Joback Method
dvisc	0.0002532	Paxs	537.99	Joback Method
dvisc	0.0001908	Paxs	582.83	Joback Method
dvisc	0.0001497	Paxs	627.67	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U292342&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U292342&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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