

# 3,4-Dihydroxy-5-methoxybenzaldehyde

<b>Other names:</b>	Benzaldehyde, 3,4-dihydroxy-5-methoxy- Protocatechualdehyde, 5-methoxy- 5-Methoxyprotocatechualdehyde 5-Hydroxyvanillin
<b>Inchi:</b>	InChI=1S/C8H8O4/c1-12-7-3-5(4-9)2-6(10)8(7)11/h2-4,10-11H,1H3
<b>InchiKey:</b>	RRKMWVISRMWBAL-UHFFFAOYSA-N
<b>Formula:</b>	C8H8O4
<b>SMILES:</b>	COc1cc(C=O)cc(O)c1O
<b>Mol. weight [g/mol]:</b>	168.15
<b>CAS:</b>	3934-87-0

## Physical Properties

Property code	Value	Unit	Source
gf	-394.50	kJ/mol	Joback Method
hf	-555.81	kJ/mol	Joback Method
hfus	25.17	kJ/mol	Joback Method
hvap	71.50	kJ/mol	Joback Method
log10ws	-0.93		Crippen Method
logp	0.919		Crippen Method
mcvol	119.000	ml/mol	McGowan Method
pc	5782.94	kPa	Joback Method
tb	646.42	K	Joback Method
tc	883.71	K	Joback Method
tf	506.53	K	Joback Method
vc	0.343	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	299.41	J/molxK	646.42	Joback Method
cpg	337.05	J/molxK	844.16	Joback Method
cpg	330.12	J/molxK	804.61	Joback Method
cpg	323.00	J/molxK	765.07	Joback Method
cpg	315.58	J/molxK	725.52	Joback Method

cpg	307.75	J/mol×K	685.97	Joback Method
cpg	343.90	J/mol×K	883.71	Joback Method
dvisc	0.0000041	Paxs	646.42	Joback Method
dvisc	0.0000060	Paxs	623.11	Joback Method
dvisc	0.0000089	Paxs	599.79	Joback Method
dvisc	0.0000137	Paxs	576.48	Joback Method
dvisc	0.0000218	Paxs	553.16	Joback Method
dvisc	0.0000362	Paxs	529.85	Joback Method
dvisc	0.0000631	Paxs	506.53	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C3934870&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C3934870&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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