

# [9]Helicene

<b>Other names:</b>	Diphenanthro[3,4-c:4'3'-g]phenanthrene
<b>Inchi:</b>	InChI=1S/C38H22/c1-3-7-31-23(5-1)9-11-25-13-15-27-17-19-29-21-22-30-20-18-28-16-1
<b>InchiKey:</b>	LKCSSGNXXHZCJW-UHFFFAOYSA-N
<b>Formula:</b>	C38H22
<b>SMILES:</b>	<chem>c1ccc2c(c1)ccc1ccc3ccc4ccc5ccc6ccc7ccc8ccccc8c7c6c5c4c3c12</chem>
<b>Mol. weight [g/mol]:</b>	478.58
<b>CAS:</b>	20495-14-1

## Physical Properties

Property code	Value	Unit	Source
gf	1167.28	kJ/mol	Joback Method
hf	857.15	kJ/mol	Joback Method
hfus	61.65	kJ/mol	Joback Method
hvap	120.21	kJ/mol	Joback Method
ie	7.07	eV	NIST Webbook
ie	7.07	eV	NIST Webbook
log10ws	-15.91		Crippen Method
logp	10.912		Crippen Method
mcvol	366.840	ml/mol	McGowan Method
pc	1392.29	kPa	Joback Method
tb	1282.22	K	Joback Method
tc	1580.19	K	Joback Method
tf	893.68	K	Joback Method
vc	1.431	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1289.69	J/molxK	1282.22	Joback Method
cpg	1332.23	J/molxK	1331.88	Joback Method
cpg	1379.99	J/molxK	1381.54	Joback Method
cpg	1433.68	J/molxK	1431.21	Joback Method
cpg	1493.99	J/molxK	1480.87	Joback Method
cpg	1561.60	J/molxK	1530.53	Joback Method

cpg	1637.22	J/mol×K	1580.19	Joback Method
dvisc	0.0104640	Paxs	893.68	Joback Method
dvisc	0.0095691	Paxs	958.44	Joback Method
dvisc	0.0088502	Paxs	1023.19	Joback Method
dvisc	0.0082618	Paxs	1087.95	Joback Method
dvisc	0.0077724	Paxs	1152.71	Joback Method
dvisc	0.0073596	Paxs	1217.46	Joback Method
dvisc	0.0070073	Paxs	1282.22	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C20495141&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C20495141&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>ie:</b>	Ionization energy
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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