

# Cyclohexane, 1,1'-hexylidenebis-

<b>Other names:</b>	1,1-Dicyclohexylhexane
<b>Inchi:</b>	InChI=1S/C18H34/c1-2-3-6-15-18(16-11-7-4-8-12-16)17-13-9-5-10-14-17/h16-18H,2-15H
<b>InchiKey:</b>	QDXQNQH DJRBQIE-UHFFFAOYSA-N
<b>Formula:</b>	C18H34
<b>SMILES:</b>	CCCCC(C1CCCCC1)C1CCCCC1
<b>Mol. weight [g/mol]:</b>	250.46
<b>CAS:</b>	55030-20-1

## Physical Properties

Property code	Value	Unit	Source
chs	-11560.00	kJ/mol	NIST Webbook
gf	147.14	kJ/mol	Joback Method
hf	-311.49	kJ/mol	Joback Method
hfus	22.52	kJ/mol	Joback Method
hvap	56.13	kJ/mol	Joback Method
log10ws	-6.42		Crippen Method
logp	6.344		Crippen Method
mcvol	242.760	ml/mol	McGowan Method
pc	1564.75	kPa	Joback Method
tb	589.00 ± 3.00	K	NIST Webbook
tc	862.73	K	Joback Method
tf	276.54 ± 0.20	K	NIST Webbook
tf	281.90 ± 0.80	K	NIST Webbook
vc	0.903	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	717.53	J/mol×K	649.90	Joback Method
cpg	836.93	J/mol×K	827.26	Joback Method
cpg	816.16	J/mol×K	791.79	Joback Method
cpg	793.89	J/mol×K	756.32	Joback Method
cpg	770.07	J/mol×K	720.84	Joback Method
cpg	744.64	J/mol×K	685.37	Joback Method

cpg	856.26	J/mol×K	862.73	Joback Method
dvisc	0.0001286	Paxs	649.90	Joback Method
dvisc	0.0001836	Paxs	590.31	Joback Method
dvisc	0.0002840	Paxs	530.73	Joback Method
dvisc	0.0004903	Paxs	471.14	Joback Method
dvisc	0.0009915	Paxs	411.55	Joback Method
dvisc	0.0025451	Paxs	351.97	Joback Method
dvisc	0.0095938	Paxs	292.38	Joback Method

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.42797e+01
Coeff. B	-4.66602e+03
Coeff. C	-1.06046e+02
Temperature range (K), min.	439.52
Temperature range (K), max.	626.33

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C55030201&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C55030201&amp;Units=SI</a>
<b>The Yaws Handbook of Vapor Pressure:</b>	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>chs:</b>	Standard solid enthalpy of combustion
<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation

<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvac:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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