

# 4-Chlorobenzoic acid, decyl ester

<b>Other names:</b>	Decyl 4-chlorobenzoate
<b>Inchi:</b>	InChI=1S/C17H25ClO2/c1-2-3-4-5-6-7-8-9-14-20-17(19)15-10-12-16(18)13-11-15/h10-13
<b>InchiKey:</b>	ASTUITNORCWJGQ-UHFFFAOYSA-N
<b>Formula:</b>	C17H25ClO2
<b>SMILES:</b>	CCCCCCCCCOCC(=O)c1ccc(Cl)cc1
<b>Mol. weight [g/mol]:</b>	296.83
<b>CAS:</b>	70136-01-5

## Physical Properties

Property code	Value	Unit	Source
gf	-50.81	kJ/mol	Joback Method
hf	-429.69	kJ/mol	Joback Method
hfus	40.42	kJ/mol	Joback Method
hvap	69.92	kJ/mol	Joback Method
log10ws	-6.17		Crippen Method
logp	5.638		Crippen Method
mcvol	246.310	ml/mol	McGowan Method
pc	1568.47	kPa	Joback Method
rinpol	2140.00		NIST Webbook
rinpol	2129.00		NIST Webbook
rinpol	2152.00		NIST Webbook
rinpol	2133.00		NIST Webbook
rinpol	2122.00		NIST Webbook
rinpol	2113.00		NIST Webbook
ripol	2682.00		NIST Webbook
ripol	2661.00		NIST Webbook
ripol	2696.00		NIST Webbook
ripol	2694.00		NIST Webbook
ripol	2672.00		NIST Webbook
ripol	2656.00		NIST Webbook
tb	733.74	K	Joback Method
tc	931.90	K	Joback Method
tf	422.37	K	Joback Method
vc	0.953	m <sup>3</sup> /kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	685.02	J/molxK	733.74	Joback Method
cpg	757.11	J/molxK	898.87	Joback Method
cpg	744.50	J/molxK	865.85	Joback Method
cpg	731.01	J/molxK	832.82	Joback Method
cpg	716.62	J/molxK	799.79	Joback Method
cpg	701.30	J/molxK	766.77	Joback Method
cpg	768.87	J/molxK	931.90	Joback Method
dvisc	0.0000974	Paxs	733.74	Joback Method
dvisc	0.0001253	Paxs	681.85	Joback Method
dvisc	0.0001681	Paxs	629.95	Joback Method
dvisc	0.0002378	Paxs	578.06	Joback Method
dvisc	0.0003602	Paxs	526.16	Joback Method
dvisc	0.0005974	Paxs	474.26	Joback Method
dvisc	0.0011220	Paxs	422.37	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C70136015&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C70136015&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient

<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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