

# 5,6-Dihydro-2-ethyl-4,6-dimethyl-4H-1,3,5-dithiazinane

<b>Other names:</b>	2-Ethyl-4,6-dimethyl-[1,3,5]dithiazinane 1,3,5-Dithiazine, perhydro-2-ethyl-4,6-dimethyl
<b>Inchi:</b>	InChI=1S/C7H15NS2/c1-4-7-9-5(2)8-6(3)10-7/h5-8H,4H2,1-3H3
<b>InchiKey:</b>	BHTMBVSWSZTUBT-UHFFFAOYSA-N
<b>Formula:</b>	C7H15NS2
<b>SMILES:</b>	CCC1SC(C)NC(C)S1
<b>Mol. weight [g/mol]:</b>	177.33

## Physical Properties

Property code	Value	Unit	Source
gf	184.52	kJ/mol	Joback Method
hf	-45.84	kJ/mol	Joback Method
hfus	24.77	kJ/mol	Joback Method
hvap	49.37	kJ/mol	Joback Method
log10ws	-3.43		Crippen Method
logp	2.484		Crippen Method
mcvol	141.310	ml/mol	McGowan Method
pc	3310.55	kPa	Joback Method
rinpol	1282.00		NIST Webbook
rinpol	1279.00		NIST Webbook
ripol	1750.00		NIST Webbook
ripol	1750.00		NIST Webbook
tb	513.98	K	Joback Method
tc	754.24	K	Joback Method
tf	439.48	K	Joback Method
vc	0.487	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	311.63	J/molxK	513.98	Joback Method
cpg	329.14	J/molxK	554.02	Joback Method
cpg	345.73	J/molxK	594.07	Joback Method
cpg	361.40	J/molxK	634.11	Joback Method

cpg	376.16	J/mol×K	674.15	Joback Method
cpg	390.00	J/mol×K	714.20	Joback Method
cpg	402.94	J/mol×K	754.24	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R62243&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R62243&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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