

# Docosanoic acid, methyl ester

<b>Other names:</b>	22:0, Me ester Behenic acid, methyl ester Kemester 9022 Methyl behenate Methyl behenoate Methyl docosanoate n-Docosanoic acid methyl ester
<b>Inchi:</b>	InChI=1S/C23H46O2/c1-3-4-5-6-7-8-9-10-11-12-13-14-15-16-17-18-19-20-21-22-23(24)
<b>InchiKey:</b>	QSQLTHHMFHEFIY-UHFFFAOYSA-N
<b>Formula:</b>	C23H46O2
<b>SMILES:</b>	CCCCCCCCCCCCCCCCCCCC(=O)OC
<b>Mol. weight [g/mol]:</b>	354.61
<b>CAS:</b>	929-77-1

## Physical Properties

Property code	Value	Unit	Source
chs	-14565.00 ± 0.40	kJ/mol	NIST Webbook
gf	-91.14	kJ/mol	Joback Method
hf	-762.85	kJ/mol	Joback Method
hfus	58.11	kJ/mol	Joback Method
hvap	126.10 ± 2.50	kJ/mol	NIST Webbook
hvap	126.00 ± 0.30	kJ/mol	NIST Webbook
hvap	126.10	kJ/mol	NIST Webbook
log10ws	-8.31		Crippen Method
logp	7.981		Crippen Method
mcvol	342.370	ml/mol	McGowan Method
pc	875.32	kPa	Joback Method
rinpol	2512.00		NIST Webbook
rinpol	2494.70		NIST Webbook
rinpol	2495.90		NIST Webbook
rinpol	2497.10		NIST Webbook
rinpol	2498.30		NIST Webbook
rinpol	2513.00		NIST Webbook
rinpol	2515.00		NIST Webbook
rinpol	2515.00		NIST Webbook
rinpol	2530.00		NIST Webbook
rinpol	2531.00		NIST Webbook

rinpol	2502.00		NIST Webbook
rinpol	2511.00		NIST Webbook
rinpol	2493.60		NIST Webbook
rinpol	2510.00		NIST Webbook
rinpol	2531.00		NIST Webbook
rinpol	2492.50		NIST Webbook
rinpol	2527.50		NIST Webbook
rinpol	2528.80		NIST Webbook
rinpol	2512.00		NIST Webbook
rinpol	2517.00		NIST Webbook
rinpol	2530.00		NIST Webbook
rinpol	2511.00		NIST Webbook
rinpol	2538.00		NIST Webbook
rinpol	2531.00		NIST Webbook
rinpol	2504.00		NIST Webbook
rinpol	2511.00		NIST Webbook
rinpol	2514.00		NIST Webbook
rinpol	2492.00		NIST Webbook
rinpol	2509.00		NIST Webbook
rinpol	2524.00		NIST Webbook
rinpol	2491.40		NIST Webbook
rinpol	419.20		NIST Webbook
rinpol	420.50		NIST Webbook
rinpol	420.60		NIST Webbook
rinpol	404.49		NIST Webbook
rinpol	2511.00		NIST Webbook
rinpol	2517.00		NIST Webbook
rinpol	2527.50		NIST Webbook
tb	801.93	K	Joback Method
tc	982.34	K	Joback Method
tf	421.13	K	Joback Method
vc	1.347	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1171.55	J/mol×K	922.21	Joback Method
cpg	1204.49	J/mol×K	982.34	Joback Method
cpg	1188.50	J/mol×K	952.27	Joback Method
cpg	1093.53	J/mol×K	801.93	Joback Method
cpg	1114.64	J/mol×K	832.00	Joback Method

cpg	1134.65	J/molxK	862.07	Joback Method
cpg	1153.61	J/molxK	892.14	Joback Method
dvisc	0.0027087	Paxs	363.15	Densities and Viscosities of Minority Fatty Acid Methyl and Ethyl Esters Present in Biodiesel
dvisc	0.0025020	Paxs	368.15	Densities and Viscosities of Minority Fatty Acid Methyl and Ethyl Esters Present in Biodiesel
dvisc	0.0023169	Paxs	373.15	Densities and Viscosities of Minority Fatty Acid Methyl and Ethyl Esters Present in Biodiesel
dvisc	0.0047493	Paxs	333.15	Densities and Viscosities of Minority Fatty Acid Methyl and Ethyl Esters Present in Biodiesel
dvisc	0.0042736	Paxs	338.15	Densities and Viscosities of Minority Fatty Acid Methyl and Ethyl Esters Present in Biodiesel
dvisc	0.0038657	Paxs	343.15	Densities and Viscosities of Minority Fatty Acid Methyl and Ethyl Esters Present in Biodiesel
dvisc	0.0035133	Paxs	348.15	Densities and Viscosities of Minority Fatty Acid Methyl and Ethyl Esters Present in Biodiesel
dvisc	0.0032066	Paxs	353.15	Densities and Viscosities of Minority Fatty Acid Methyl and Ethyl Esters Present in Biodiesel

dvisc	0.0029392	Paxs	358.15	Densities and Viscosities of Minority Fatty Acid Methyl and Ethyl Esters Present in Biodiesel
hfust	82.30	kJ/mol	327.20	NIST Webbook
hfust	83.50	kJ/mol	327.20	NIST Webbook
hfust	82.30	kJ/mol	325.00	NIST Webbook
hfust	29.20	kJ/mol	324.99	NIST Webbook
hvapt	98.20	kJ/mol	503.00	NIST Webbook
hvapt	81.00	kJ/mol	498.00	NIST Webbook
hvapt	127.50	kJ/mol	298.15	the vaporization enthalpies and vapor pressures of a series of unstaured fatty acid methyl esters by correlation gas chromatography

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	2.53807e+01
Coeff. B	-1.04203e+04
Coeff. C	-1.59063e+02
Temperature range (K), min.	574.33
Temperature range (K), max.	678.28

## Sources

The Yaws Handbook of Vapor Pressure:

Phase behaviour of high molecular mass methyl esters in supercritical fluid  
 the vaporization enthalpies and vapor pressures of a series of unstaured fatty acid methyl esters by correlation gas chromatography  
 Densities and Viscosities of Minority Fatty Acid Methyl and Ethyl Esters Present in Biodiesel:

NIST Webbook:

Crippen Method:

<https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure>

<https://www.doi.org/10.1016/j.fluid.2011.08.015>

<https://www.doi.org/10.1016/j.tca.2007.02.008>

<https://www.doi.org/10.1021/je1012235>

[https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)

<http://webbook.nist.gov/cgi/cbook.cgi?ID=C929771&Units=SI>

<http://pubs.acs.org/doi/abs/10.1021/ci9903071>

**Crippen Method:** [https://www.chemeo.com/doc/models/crippen\\_log10ws](https://www.chemeo.com/doc/models/crippen_log10ws)

**McGowan Method:** <http://link.springer.com/article/10.1007/BF02311772>

## Legend

<b>chs:</b>	Standard solid enthalpy of combustion
<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hfust:</b>	Enthalpy of fusion at a given temperature
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>hvapt:</b>	Enthalpy of vaporization at a given temperature
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>rinpola:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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