2-Pyrrolidinone, 1-ethenyl-

Other names: 1-Ethenyl-2-pyrrolidinone

1-Vinyl-2-pyrrolidinone

1-Vinyl-2-pyrrolidinone, monomer

1-Vinyl-2-pyrrolidone1-Vinylpyrrolidinone1-Vinylpyrrolidone

2-Pyrrolidinone, 1-vinyl-N-Vinyl-2-pyrrolidinone N-Vinyl-2-pyrrolidinone N-Vinylpyrrolidinone N-vinylpyrrolidinone

NSC 10222 V-Pyrol

Vinyl-2-pyrrolidone Vinylbutyrolactam Vinylpyrrolidinone Vinylpyrrolidone pyrrolidone, N-vinyl-

InChl=1S/C6H9NO/c1-2-7-5-3-4-6(7)8/h2H,1,3-5H2

InchiKey: WHNWPMSKXPGLAX-UHFFFAOYSA-N

Formula: C6H9NO

SMILES: C=CN1CCCC1=O

Mol. weight [g/mol]: 111.14 **CAS:** 88-12-0

Physical Properties

Property code	Value	Unit	Source
log10ws	-0.92		Crippen Method
logp	0.752		Crippen Method
mcvol	91.790	ml/mol	McGowan Method
rinpol	1102.00		NIST Webbook
rinpol	1102.00		NIST Webbook
rinpol	1077.00		NIST Webbook
rinpol	1102.00		NIST Webbook

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K	[] Source	
dvisc	0.0043220	Paxs		Densities, Viscosities, Speeds of Sound, and Relative Permittivities for Water + Cyclic Amides (2-Pyrrolidinone, Methyl-2-pyrrolidinor and -Vinyl-2-pyrrolidinone at Different Temperatures	
dvisc	0.0029420	Paxs		Densities, Viscosities, Speeds of Sound, and Relative Permittivities for Water + Cyclic Amides (2-Pyrrolidinone, Methyl-2-pyrrolidinor and -Vinyl-2-pyrrolidinone at Different Temperatures	
dvisc	0.0025900	Paxs		Densities, Viscosities, Speeds of Sound, and Relative Permittivities for Water + Cyclic Amides (2-Pyrrolidinone, Methyl-2-pyrrolidinone at Different Temperatures	
dvisc	0.0022590	Paxs		Densities, Viscosities, Speeds of Sound, and Relative Permittivities for Water + Cyclic Amides (2-Pyrrolidinone, Methyl-2-pyrrolidinor and -Vinyl-2-pyrrolidinone at Different Temperatures	

dvisc	0.0018360	Paxs		Densities, Viscosities, Speeds of Sound, and Relative Permittivities for Water + Cyclic Amides (2-Pyrrolidinone, -Methyl-2-pyrrolidinone, and I-Vinyl-2-pyrrolidinone) at Different Temperatures
hfust	15.28	kJ/mol	286.20	NIST Webbook
hfust	15.28	kJ/mol	286.20	NIST Webbook
rhol	1039.36	kg/m3		Density, Refractive Index, and Speed of Sound of the Binary Mixture of Butyl-3-methylimidazolium Tetrafluoroborate +
			<u> </u>	N-Vinyl-2-pyrrolidinone from T = (298.15 to 323.15) K at Atmospheric Pressure
rhol	1034.95	kg/m3		Density, Refractive Index, and Speed of Sound of the Binary Mixture of Butyl-3-methylimidazolium Tetrafluoroborate + N-Vinyl-2-pyrrolidinone from T = (298.15 to 323.15) K at Atmospheric Pressure
rhol	1030.55	kg/m3		Density, Refractive Index, and Speed of Sound of the Binary Mixture of Butyl-3-methylimidazolium Tetrafluoroborate + N-Vinyl-2-pyrrolidinone from T = (298.15 to 323.15) K at Atmospheric Pressure

rhol	1026.14	kg/m3	313.15 Density, Refractive Index, and Speed of Sound of the Binary Mixture of 1-Butyl-3-methylimidazolium Tetrafluoroborate + N-Vinyl-2-pyrrolidinone from T = (298.15 to 323.15) K at Atmospheric Pressure
rhol	1021.73	kg/m3	318.15 Density, Refractive Index, and Speed of Sound of the Binary Mixture of 1-Butyl-3-methylimidazolium Tetrafluoroborate + N-Vinyl-2-pyrrolidinone from T = (298.15 to 323.15) K at Atmospheric Pressure
rhol	1017.33	kg/m3	323.15 Density, Refractive Index, and Speed of Sound of the Binary Mixture of 1-Butyl-3-methylimidazolium Tetrafluoroborate + N-Vinyl-2-pyrrolidinone from T = (298.15 to 323.15) K at Atmospheric Pressure

Sources

Crippen Method: http://pubs.acs.org/doi/abs/10.1021/ci990307l

Crippen Method: https://www.chemeo.com/doc/models/crippen_log10ws

Molecular interaction between binary Molecular interaction between binary mixtures 1-butyl-3-methylimidazolium bildufility of many length of the properties, Speeds of Sound, and Relative Permittivities for Density, Coffic hind legex, and Speed of Sound and Relative Permittivities for Density, Coffic hind legex, and Speed of Sound and Relative Permittivities for Density, Coffic hind legex, and Speed of Sound and Relative Permittivities for Density, Coffic hind legex, and Speed of Tattally 1995 find leavel, land Tattally 1995 find leavel, land

Pressure:

https://www.doi.org/10.1016/j.jct.2017.01.014

https://www.doi.org/10.1021/acs.jced.8b00126

http://link.springer.com/article/10.1007/BF02311772

http://webbook.nist.gov/cgi/cbook.cgi?ID=C88120&Units=SI

Legend

dvisc: Dynamic viscosity

hfust: Enthalpy of fusion at a given temperature

log10ws:Log10 of Water solubility in mol/llogp:Octanol/Water partition coefficientmcvol:McGowan's characteristic volume

rhol: Liquid Density

rinpol: Non-polar retention indices

Latest version available from:

https://www.chemeo.com/cid/45-376-5/2-Pyrrolidinone-1-ethenyl.pdf

Generated by Cheméo on 2025-12-05 21:53:14.450426507 +0000 UTC m=+4719791.980467172.

Cheméo (https://www.chemeo.com) is the biggest free database of chemical and physical data for the process industry.