

Benzene, 1-(2,2-dimethyl-1-methylenepropyl)-3-fluoro-

Inchi: InChI=1S/C12H15F/c1-9(12(2,3)4)10-6-5-7-11(13)8-10/h5-8H,1H2,2-4H3
InchiKey: AGFFWVOXOGOEID-UHFFFAOYSA-N
Formula: C12H15F
SMILES: C=C(c1cccc(F)c1)C(C)(C)C
Mol. weight [g/mol]: 178.25
CAS: 146558-44-3

Physical Properties

Property code	Value	Unit	Source
affp	838.80	kJ/mol	NIST Webbook
basg	809.90	kJ/mol	NIST Webbook
gf	40.26	kJ/mol	Joback Method
hf	-155.17	kJ/mol	Joback Method
hfus	13.56	kJ/mol	Joback Method
hvap	42.54	kJ/mol	Joback Method
log10ws	-4.06		Crippen Method
logp	3.885		Crippen Method
mcvol	153.650	ml/mol	McGowan Method
pc	2419.50	kPa	Joback Method
tb	498.22	K	Joback Method
tc	709.15	K	Joback Method
tf	251.23	K	Joback Method
vc	0.589	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	338.89	J/molxK	498.22	Joback Method
cpg	355.54	J/molxK	533.37	Joback Method
cpg	371.11	J/molxK	568.53	Joback Method
cpg	385.64	J/molxK	603.68	Joback Method
cpg	399.21	J/molxK	638.84	Joback Method
cpg	411.87	J/molxK	673.99	Joback Method
cpg	423.67	J/molxK	709.15	Joback Method

Sources

Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=C146558443&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307I
Crippen Method:	https://www.cheméo.com/doc/models/crippen_log10ws

Legend

affp:	Proton affinity
basg:	Gas basicity
cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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