

# Phenol, 3,4,5-trimethoxy-

<b>Other names:</b>	3,4,5-Trimethoxyphenol Antiarol
<b>Inchi:</b>	InChI=1S/C9H12O4/c1-11-7-4-6(10)5-8(12-2)9(7)13-3/h4-5,10H,1-3H3
<b>InchiKey:</b>	VTCDZPUMZAZMSB-UHFFFAOYSA-N
<b>Formula:</b>	C9H12O4
<b>SMILES:</b>	COc1cc(O)cc(OC)c1OC
<b>Mol. weight [g/mol]:</b>	184.19
<b>CAS:</b>	642-71-7

## Physical Properties

Property code	Value	Unit	Source
gf	-351.57	kJ/mol	Joback Method
hf	-589.47	kJ/mol	Joback Method
hfus	31.94	kJ/mol	3,4,5-Trimethoxyphenol: A combined experimental and theoretical thermochemical investigation of its antioxidant capacity
hvap	59.47	kJ/mol	Joback Method
log10ws	-1.38		Crippen Method
logp	1.418		Crippen Method
mcvol	137.390	ml/mol	McGowan Method
pc	3530.46	kPa	Joback Method
ripol	1599.00		NIST Webbook
ripol	1614.00		NIST Webbook
ripol	3026.00		NIST Webbook
ripol	3065.00		NIST Webbook
ripol	3060.00		NIST Webbook
ripol	3060.00		NIST Webbook
ripol	3049.00		NIST Webbook
ripol	3044.00		NIST Webbook
ripol	3049.00		NIST Webbook
ripol	3060.00		NIST Webbook
tb	589.84	K	Joback Method
tc	806.43	K	Joback Method
tf	421.06	K	Joback Method
vc	0.452	m <sup>3</sup> /kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	396.73	J/molxK	806.43	Joback Method
cpg	334.87	J/molxK	589.84	Joback Method
cpg	346.51	J/molxK	625.94	Joback Method
cpg	357.61	J/molxK	662.04	Joback Method
cpg	368.17	J/molxK	698.13	Joback Method
cpg	378.20	J/molxK	734.23	Joback Method
cpg	387.72	J/molxK	770.33	Joback Method
dvisc	0.0000238	Paxs	589.84	Joback Method
dvisc	0.0003535	Paxs	421.06	Joback Method
dvisc	0.0001959	Paxs	449.19	Joback Method
dvisc	0.0001164	Paxs	477.32	Joback Method
dvisc	0.0000733	Paxs	505.45	Joback Method
dvisc	0.0000485	Paxs	533.58	Joback Method
dvisc	0.0000334	Paxs	561.71	Joback Method
hfust	31.94	kJ/mol	420.20	NIST Webbook

## Sources

McGowan Method:	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
NIST Webbook:	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C642717&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C642717&amp;Units=SI</a>
Crippen Method:	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
Crippen Method:	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
3,4,5-Trimethoxyphenol: A combined experimental and theoretical thermodynamic investigation of its antioxidant capacity:	<a href="https://www.doi.org/10.1016/j.jct.2007.11.006">https://www.doi.org/10.1016/j.jct.2007.11.006</a>
Joback Method	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

cpg:	Ideal gas heat capacity
dvisc:	Dynamic viscosity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions

<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hfust:</b>	Enthalpy of fusion at a given temperature
<b>hvac:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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