

# 3-Hexene, 3-methyl-, (E)-

<b>Other names:</b>	(E)-3-Hexene, 3-methyl (E)-3-METHYL-3-HEXENE 3-METHYL-TRANS-3-HEXENE 3-Methyl-3-hexene, trans TRANS-3-METHYL-3-HEXENE
<b>Inchi:</b>	InChI=1S/C7H14/c1-4-6-7(3)5-2/h6H,4-5H2,1-3H3/b7-6+
<b>InchiKey:</b>	FHHSSXNRVNXTBG-VOTSOKGWSA-N
<b>Formula:</b>	C7H14
<b>SMILES:</b>	CCC=C(C)CC
<b>Mol. weight [g/mol]:</b>	98.19
<b>CAS:</b>	3899-36-3

## Physical Properties

Property code	Value	Unit	Source
chl	-4642.65 ± 0.88	kJ/mol	NIST Webbook
gf	79.73	kJ/mol	Joback Method
hf	-80.38	kJ/mol	Joback Method
hfus	12.78	kJ/mol	Joback Method
hvap	35.80	kJ/mol	NIST Webbook
hvap	35.80	kJ/mol	NIST Webbook
log10ws	-2.61		Crippen Method
logp	2.753		Crippen Method
mcvol	105.190	ml/mol	McGowan Method
pc	2976.29	kPa	Joback Method
rinpol	703.40		NIST Webbook
rinpol	691.20		NIST Webbook
rinpol	703.00		NIST Webbook
rinpol	703.00		NIST Webbook
rinpol	691.00		NIST Webbook
rinpol	692.00		NIST Webbook
rinpol	691.00		NIST Webbook
rinpol	691.00		NIST Webbook
rinpol	689.00		NIST Webbook
rinpol	685.00		NIST Webbook
rinpol	685.00		NIST Webbook
rinpol	686.00		NIST Webbook
rinpol	686.00		NIST Webbook

rinpol	695.00		NIST Webbook
rinpol	684.00		NIST Webbook
rinpol	684.00		NIST Webbook
rinpol	695.00		NIST Webbook
rinpol	703.20		NIST Webbook
rinpol	703.20		NIST Webbook
rinpol	693.00		NIST Webbook
rinpol	691.30		NIST Webbook
rinpol	691.00		NIST Webbook
rinpol	685.00		NIST Webbook
rinpol	695.00		NIST Webbook
rinpol	698.00		NIST Webbook
rinpol	703.00		NIST Webbook
rinpol	703.00		NIST Webbook
rinpol	685.00		NIST Webbook
rinpol	703.00		NIST Webbook
ripol	735.00		NIST Webbook
ripol	731.00		NIST Webbook
tb	366.70 ± 0.30	K	NIST Webbook
tb	366.67 ± 0.20	K	NIST Webbook
tb	366.70	K	NIST Webbook
tc	539.41	K	Joback Method
tf	131.59 ± 0.03	K	NIST Webbook
vc	0.408	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	224.86	J/mol×K	480.80	Joback Method
cpg	234.79	J/mol×K	510.11	Joback Method
cpg	180.34	J/mol×K	363.60	Joback Method
cpg	192.22	J/mol×K	392.90	Joback Method
cpg	203.59	J/mol×K	422.20	Joback Method
cpg	214.47	J/mol×K	451.50	Joback Method
cpg	244.29	J/mol×K	539.41	Joback Method
hvapt	34.80	kJ/mol	339.00	NIST Webbook
hvapt	35.30	kJ/mol	333.50	NIST Webbook

# Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.41951e+01
Coeff. B	-3.11299e+03
Coeff. C	-4.16350e+01
Temperature range (K), min.	265.47
Temperature range (K), max.	392.05

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/T + C \cdot \ln(T) + D \cdot T^2$
Coeff. A	8.98033e+01
Coeff. B	-7.19630e+03
Coeff. C	-1.13231e+01
Coeff. D	9.64776e-06
Temperature range (K), min.	300.15
Temperature range (K), max.	366.15

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>KDB:</b>	<a href="https://www.thermo.com/research/kdb/hcprop/showprop.php?cmpid=232">https://www.thermo.com/research/kdb/hcprop/showprop.php?cmpid=232</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C3899363&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C3899363&amp;Units=SI</a>
<b>The Yaws Handbook of Vapor Pressure:</b>	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>
<b>KDB Vapor Pressure Data:</b>	<a href="https://www.thermo.com/research/kdb/hcprop/showprop.php?cmpid=232">https://www.thermo.com/research/kdb/hcprop/showprop.php?cmpid=232</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemo.com/doc/models/crippen_log10ws">https://www.chemo.com/doc/models/crippen_log10ws</a>

## Legend

chl: Standard liquid enthalpy of combustion

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>h vap:</b>	Enthalpy of vaporization at standard conditions
<b>hvapt:</b>	Enthalpy of vaporization at a given temperature
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/45-528-6/3-Hexene-3-methyl-E.pdf>

Generated by Cheméo on 2024-04-27 03:15:29.942655724 +0000 UTC m=+16476978.863233040.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.