

Undecane, 3,8-dimethyl-

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|-----------------------------|---|
| Other names: | 3,8-Dimethylundecane |
| Inchi: | InChI=1S/C13H28/c1-5-9-13(4)11-8-7-10-12(3)6-2/h12-13H,5-11H2,1-4H3 |
| InchiKey: | WOGVWUCXENBLIV-UHFFFAOYSA-N |
| Formula: | C13H28 |
| SMILES: | CCCC(C)CCCC(C)CC |
| Mol. weight [g/mol]: | 184.36 |
| CAS: | 17301-30-3 |

Physical Properties

| Property code | Value | Unit | Source |
|---------------|---------|---------|----------------|
| gf | 53.70 | kJ/mol | Joback Method |
| hf | -322.21 | kJ/mol | Joback Method |
| hfus | 22.38 | kJ/mol | Joback Method |
| hvap | 43.76 | kJ/mol | Joback Method |
| log10ws | -4.78 | | Crippen Method |
| logp | 5.029 | | Crippen Method |
| mcvol | 194.030 | ml/mol | McGowan Method |
| pc | 1665.97 | kPa | Joback Method |
| rinpol | 1227.00 | | NIST Webbook |
| rinpol | 1228.00 | | NIST Webbook |
| rinpol | 1227.00 | | NIST Webbook |
| rinpol | 1228.00 | | NIST Webbook |
| tb | 495.96 | K | Joback Method |
| tc | 662.44 | K | Joback Method |
| tf | 206.27 | K | Joback Method |
| vc | 0.751 | m3/kmol | Joback Method |

Temperature Dependent Properties

| Property code | Value | Unit | Temperature [K] | Source |
|---------------|--------|---------|-----------------|---------------|
| cpg | 455.70 | J/molxK | 495.96 | Joback Method |
| cpg | 540.06 | J/molxK | 634.69 | Joback Method |
| cpg | 524.55 | J/molxK | 606.95 | Joback Method |
| cpg | 508.38 | J/molxK | 579.20 | Joback Method |

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|-------|-----------|---------|--------|---------------|
| cpg | 491.53 | J/mol×K | 551.45 | Joback Method |
| cpg | 473.97 | J/mol×K | 523.71 | Joback Method |
| cpg | 554.93 | J/mol×K | 662.44 | Joback Method |
| dvisc | 0.0001780 | Paxs | 495.96 | Joback Method |
| dvisc | 0.0002527 | Paxs | 447.68 | Joback Method |
| dvisc | 0.0003903 | Paxs | 399.40 | Joback Method |
| dvisc | 0.0006795 | Paxs | 351.12 | Joback Method |
| dvisc | 0.0014115 | Paxs | 302.83 | Joback Method |
| dvisc | 0.0038696 | Paxs | 254.55 | Joback Method |
| dvisc | 0.0170093 | Paxs | 206.27 | Joback Method |

Sources

| | |
|------------------------|---|
| Joback Method: | https://en.wikipedia.org/wiki/Joback_method |
| McGowan Method: | http://link.springer.com/article/10.1007/BF02311772 |
| NIST Webbook: | http://webbook.nist.gov/cgi/cbook.cgi?ID=C17301303&Units=SI |
| Crippen Method: | http://pubs.acs.org/doi/abs/10.1021/ci990307l |
| Crippen Method: | https://www.chemeo.com/doc/models/crippen_log10ws |

Legend

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|-----------------|---|
| cpg: | Ideal gas heat capacity |
| dvisc: | Dynamic viscosity |
| gf: | Standard Gibbs free energy of formation |
| hf: | Enthalpy of formation at standard conditions |
| hfus: | Enthalpy of fusion at standard conditions |
| hvap: | Enthalpy of vaporization at standard conditions |
| log10ws: | Log10 of Water solubility in mol/l |
| logp: | Octanol/Water partition coefficient |
| mcvol: | McGowan's characteristic volume |
| pc: | Critical Pressure |
| rinpol: | Non-polar retention indices |
| tb: | Normal Boiling Point Temperature |
| tc: | Critical Temperature |
| tf: | Normal melting (fusion) point |
| vc: | Critical Volume |

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