

# «alpha»-Campholenol

<b>Other names:</b>	«alpha»-Campholene alcohol 2,2,3-Trimethyl-cyclopent-3-ene,1-ethanol campholene alcohol
<b>Inchi:</b>	InChI=1S/C10H18O/c1-8-4-5-9(6-7-11)10(8,2)3/h4,9,11H,5-7H2,1-3H3
<b>InchiKey:</b>	NPGPPCSBEMHHCR-UHFFFAOYSA-N
<b>Formula:</b>	C10H18O
<b>SMILES:</b>	CC1=CCC(CCO)C1(C)C
<b>Mol. weight [g/mol]:</b>	154.25
<b>CAS:</b>	1901-38-8

## Physical Properties

Property code	Value	Unit	Source
gf	-59.82	kJ/mol	Joback Method
hf	-300.27	kJ/mol	Joback Method
hfus	15.28	kJ/mol	Joback Method
hvap	54.28	kJ/mol	Joback Method
log10ws	-2.54		Crippen Method
logp	2.361		Crippen Method
mcvol	142.470	ml/mol	McGowan Method
pc	2850.52	kPa	Joback Method
ripol	1185.00		NIST Webbook
ripol	1185.00		NIST Webbook
ripol	1186.00		NIST Webbook
ripol	1202.00		NIST Webbook
ripol	1790.00		NIST Webbook
ripol	1805.00		NIST Webbook
ripol	1805.00		NIST Webbook
ripol	1782.00		NIST Webbook
ripol	1800.00		NIST Webbook
ripol	1800.00		NIST Webbook
ripol	1782.00		NIST Webbook
ripol	1790.00		NIST Webbook
tb	535.37	K	Joback Method
tc	724.71	K	Joback Method
tf	307.12	K	Joback Method
vc	0.538	m <sup>3</sup> /kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	350.35	J/mol×K	535.37	Joback Method
cpg	364.68	J/mol×K	566.93	Joback Method
cpg	378.25	J/mol×K	598.48	Joback Method
cpg	391.13	J/mol×K	630.04	Joback Method
cpg	403.40	J/mol×K	661.59	Joback Method
cpg	415.13	J/mol×K	693.15	Joback Method
cpg	426.40	J/mol×K	724.71	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C1901388&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C1901388&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature

**tf:** Normal melting (fusion) point

**vc:** Critical Volume

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