

# Benzene, 1-methoxy-2-(2-nitroethenyl)-

<b>Other names:</b>	1-Methoxy-2-(2-nitroethenyl)benzene 1-(2-Methoxyphenyl)-2-nitroethene 2-(2-Nitrovinyl)anisole 2-Methoxy-«beta»-nitrostyrene o-Methoxy-«beta»-nitrostyrene Anisole, o-(2-nitrovinyl)- 1-Methoxy-2-(2-nitrovinyl)benzene Ethene,-1-(2-methoxyphenyl)-2-nitro- NSC 170699 NSC 59450
<b>Inchi:</b>	InChI=1S/C9H9NO3/c1-13-9-5-3-2-4-8(9)6-7-10(11)12/h2-7H,1H3/b7-6+
<b>InchiKey:</b>	FVKSRNVYJXQCLK-VOTSOKGWSA-N
<b>Formula:</b>	C9H9NO3
<b>SMILES:</b>	<chem>COc1ccccc1C=C[N+](=O)[O-]</chem>
<b>Mol. weight [g/mol]:</b>	179.17
<b>CAS:</b>	3316-24-3

## Physical Properties

Property code	Value	Unit	Source
gf	138.45	kJ/mol	Joback Method
hf	-29.79	kJ/mol	Joback Method
hfus	25.47	kJ/mol	Joback Method
hvap	57.53	kJ/mol	Joback Method
log10ws	-2.99		Crippen Method
logp	1.943		Crippen Method
mcvol	132.900	ml/mol	McGowan Method
pc	3384.14	kPa	Joback Method
tb	615.40	K	Joback Method
tc	861.96	K	Joback Method
tf	390.89	K	Joback Method
vc	0.511	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	312.97	J/mol×K	615.40	Joback Method
cpg	325.17	J/mol×K	656.49	Joback Method
cpg	336.46	J/mol×K	697.59	Joback Method
cpg	346.90	J/mol×K	738.68	Joback Method
cpg	356.53	J/mol×K	779.77	Joback Method
cpg	365.40	J/mol×K	820.86	Joback Method
cpg	373.54	J/mol×K	861.96	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C3316243&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C3316243&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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