

D-Alanine, N-(4-fluoro-2-trifluoromethylbenzoyl)-, undecyl ester

InChI: InChI=1S/C22H31F4NO3/c1-3-4-5-6-7-8-9-10-11-14-30-21(29)16(2)27-20(28)18-13-12-11
InChIKey: FZAQEVCLHCIBFY-UHFFFAOYSA-N

Formula: C22H31F4NO3

SMILES: CCCCCCCCCCOC(=O)C(C)NC(=O)c1ccc(F)cc1C(F)(F)F

Mol. weight [g/mol]: 433.48

Physical Properties

Property code	Value	Unit	Source
gf	-824.78	kJ/mol	Joback Method
hf	-1386.20	kJ/mol	Joback Method
hfus	56.87	kJ/mol	Joback Method
hvap	85.55	kJ/mol	Joback Method
log10ws	-7.67		Crippen Method
logp	6.037		Crippen Method
mvol	323.150	ml/mol	McGowan Method
pc	1067.97	kPa	Joback Method
rinpol	2538.00		NIST Webbook
rinpol	2538.00		NIST Webbook
tb	913.14	K	Joback Method
tc	1118.03	K	Joback Method
tf	553.69	K	Joback Method
vc	1.280	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1056.69	J/molxK	913.14	Joback Method
cpg	1071.99	J/molxK	947.29	Joback Method
cpg	1086.17	J/molxK	981.44	Joback Method
cpg	1099.31	J/molxK	1015.58	Joback Method
cpg	1111.44	J/molxK	1049.73	Joback Method
cpg	1122.65	J/molxK	1083.88	Joback Method
cpg	1132.99	J/molxK	1118.03	Joback Method

Sources

NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=U348408&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci9903071
Crippen Method:	https://www.cheméo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvp:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
rinp:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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