

# 1,9-Decadiene, 5-methyl

<b>Inchi:</b>	InChI=1S/C11H20/c1-4-6-8-10-11(3)9-7-5-2/h4-5,11H,1-2,6-10H2,3H3
<b>InchiKey:</b>	CCFMEFPUMIIUQB-UHFFFAOYSA-N
<b>Formula:</b>	C11H20
<b>SMILES:</b>	C=CCCCC(C)CCC=C
<b>Mol. weight [g/mol]:</b>	152.28

## Physical Properties

Property code	Value	Unit	Source
gf	214.98	kJ/mol	Joback Method
hf	-24.79	kJ/mol	Joback Method
hfus	18.16	kJ/mol	Joback Method
hvap	38.35	kJ/mol	Joback Method
log10ws	-3.89		Crippen Method
logp	3.945		Crippen Method
mcvol	157.250	ml/mol	McGowan Method
pc	2100.34	kPa	Joback Method
rinpol	1023.00		NIST Webbook
rinpol	1023.00		NIST Webbook
tb	444.00	K	Joback Method
tc	615.59	K	Joback Method
tf	195.21	K	Joback Method
vc	0.608	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	326.31	J/mol×K	444.00	Joback Method
cpg	397.14	J/mol×K	586.99	Joback Method
cpg	384.21	J/mol×K	558.39	Joback Method
cpg	370.68	J/mol×K	529.79	Joback Method
cpg	356.54	J/mol×K	501.20	Joback Method
cpg	341.75	J/mol×K	472.60	Joback Method
cpg	409.49	J/mol×K	615.59	Joback Method
dvisc	0.0002157	Paxs	444.00	Joback Method

dvisc	0.0002885	Paxs	402.53	Joback Method
dvisc	0.0004126	Paxs	361.07	Joback Method
dvisc	0.0006474	Paxs	319.61	Joback Method
dvisc	0.0011617	Paxs	278.14	Joback Method
dvisc	0.0025590	Paxs	236.67	Joback Method
dvisc	0.0078837	Paxs	195.21	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R2003&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R2003&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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