

# Octadecane, 5,14-dibutyl-

<b>Other names:</b>	5,14-Di-n-butyloctadecane 5,14-Dibutyloctadecane
<b>Inchi:</b>	InChI=1S/C26H54/c1-5-9-19-25(20-10-6-2)23-17-15-13-14-16-18-24-26(21-11-7-3)22-12
<b>InchiKey:</b>	VZZCEMBCJSFMMD-UHFFFAOYSA-N
<b>Formula:</b>	C26H54
<b>SMILES:</b>	CCCC(CCCC)CCCCCCCC(CCCC)CCCC
<b>Mol. weight [g/mol]:</b>	366.71
<b>CAS:</b>	55282-13-8

## Physical Properties

Property code	Value	Unit	Source
gf	163.16	kJ/mol	Joback Method
hf	-590.53	kJ/mol	Joback Method
hfus	56.05	kJ/mol	Joback Method
hvap	72.69	kJ/mol	Joback Method
log10ws	-10.22		Crippen Method
logp	10.100		Crippen Method
mvol	377.200	ml/mol	McGowan Method
pc	731.25	kPa	Joback Method
tb	793.40	K	Joback Method
tc	971.62	K	Joback Method
tf	278.90 ± 0.50	K	NIST Webbook
tf	278.90 ± 0.50	K	NIST Webbook
tf	278.85	K	NIST Webbook
tf	278.90 ± 2.00	K	NIST Webbook
vc	1.480	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1344.80	J/mol×K	971.62	Joback Method
cpg	1217.07	J/mol×K	793.40	Joback Method
cpg	1241.13	J/mol×K	823.10	Joback Method
cpg	1264.02	J/mol×K	852.81	Joback Method

cpg	1285.78	J/mol×K	882.51	Joback Method
cpg	1306.47	J/mol×K	912.22	Joback Method
cpg	1326.13	J/mol×K	941.92	Joback Method
dvisc	0.0000369	Paxs	793.40	Joback Method
dvisc	0.0031975	Paxs	352.78	Joback Method
dvisc	0.0008010	Paxs	426.22	Joback Method
dvisc	0.0003014	Paxs	499.65	Joback Method
dvisc	0.0001457	Paxs	573.09	Joback Method
dvisc	0.0000831	Paxs	646.53	Joback Method
dvisc	0.0000531	Paxs	719.96	Joback Method
hvapt	89.30	kJ/mol	483.00	NIST Webbook

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.63784e+01
Coeff. B	-6.33671e+03
Coeff. C	-1.30190e+02
Temperature range (K), min.	524.00
Temperature range (K), max.	702.77

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C55282138&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C55282138&amp;Units=SI</a>
<b>The Yaws Handbook of Vapor Pressure:</b>	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity

<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvac:</b>	Enthalpy of vaporization at standard conditions
<b>hvapt:</b>	Enthalpy of vaporization at a given temperature
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mccol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvac:</b>	Vapor pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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