

# 3-Cyclohexene-1-carboxaldehyde

<b>Other names:</b>	Cyclohexene-4-carboxaldehyde 1-Formyl-3-cyclohexene 1,2,3,6-Tetrahydrobenzaldehyde 1,2,5,6-Tetrahydrobenzaldehyde 3-Cyclohexen-1-aldehyde 4-Formylcyclohexene 1,2,3,6-Tetrahydrobenzaldehyde(1,2,5,6) 1-Cyclohexene-4-carboxaldehyde 4-Cyclohexene-1-carboxaldehyde UN 2498 NSC 16241 cyclohex-3-ene-1-carbaldehyde
<b>Inchi:</b>	InChI=1S/C7H10O/c8-6-7-4-2-1-3-5-7/h1-2,6-7H,3-5H2
<b>InchiKey:</b>	DCFDVJPDXYGCOK-UHFFFAOYSA-N
<b>Formula:</b>	C7H10O
<b>SMILES:</b>	O=CC1CC=CCC1
<b>Mol. weight [g/mol]:</b>	110.15
<b>CAS:</b>	100-50-5

## Physical Properties

Property code	Value	Unit	Source
gf	-37.05	kJ/mol	Joback Method
hf	-161.29	kJ/mol	Joback Method
hfus	9.23	kJ/mol	Joback Method
hvap	38.62	kJ/mol	Joback Method
log10ws	-1.54		Crippen Method
logp	1.542		Crippen Method
mcvol	95.900	ml/mol	McGowan Method
pc	4031.24	kPa	Joback Method
ripol	1490.00		NIST Webbook
tb	437.00 ± 3.00	K	NIST Webbook
tb	438.15	K	NIST Webbook
tb	437.20	K	NIST Webbook
tc	639.30	K	Joback Method
tf	177.10 ± 1.00	K	NIST Webbook
vc	0.363	m <sup>3</sup> /kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	183.25	J/molxK	426.93	Joback Method
cpg	196.95	J/molxK	462.32	Joback Method
cpg	209.90	J/molxK	497.72	Joback Method
cpg	222.11	J/molxK	533.11	Joback Method
cpg	233.61	J/molxK	568.51	Joback Method
cpg	244.42	J/molxK	603.90	Joback Method
cpg	254.56	J/molxK	639.30	Joback Method
dvisc	0.0049031	Paxs	218.79	Joback Method
dvisc	0.0023281	Paxs	253.48	Joback Method
dvisc	0.0013226	Paxs	288.17	Joback Method
dvisc	0.0008484	Paxs	322.86	Joback Method
dvisc	0.0005932	Paxs	357.55	Joback Method
dvisc	0.0004419	Paxs	392.24	Joback Method
dvisc	0.0003453	Paxs	426.93	Joback Method

# Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	325.20	K	1.70	NIST Webbook

## Sources

**Crippen Method:**

[https://www.chemeo.com/doc/models/crippen\\_log10ws](https://www.chemeo.com/doc/models/crippen_log10ws)

**Joback Method:**

[https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)

**McGowan Method:**

<http://link.springer.com/article/10.1007/BF02311772>

**NIST Webbook:**

<http://webbook.nist.gov/cgi/cbook.cgi?ID=C100505&Units=SI>

**Crippen Method:**

<http://pubs.acs.org/doi/abs/10.1021/ci990307l>

# Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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