

# 1-Propoxypropan-2-yl 3,5,5-trimethylhexanoate

Inchi:	InChI=1S/C15H30O3/c1-7-8-17-11-13(3)18-14(16)9-12(2)10-15(4,5)6/h12-13H,7-11H2,1
InchiKey:	UWUZSZNNQJCARS-UHFFFAOYSA-N
Formula:	C15H30O3
SMILES:	CCCOCC(C)OC(=O)CC(C)CC(C)(C)C
Mol. weight [g/mol]:	258.40

## Physical Properties

Property code	Value	Unit	Source
gf	-265.54	kJ/mol	Joback Method
hf	-749.26	kJ/mol	Joback Method
hfus	24.12	kJ/mol	Joback Method
hvap	58.48	kJ/mol	Joback Method
log10ws	-3.68		Crippen Method
logp	3.807		Crippen Method
mvol	235.520	ml/mol	McGowan Method
pc	1486.14	kPa	Joback Method
rinpol	1521.00		NIST Webbook
rinpol	1521.00		NIST Webbook
tb	637.20	K	Joback Method
tc	817.10	K	Joback Method
tf	325.62	K	Joback Method
vc	0.894	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	654.23	J/molxK	637.20	Joback Method
cpg	672.61	J/molxK	667.18	Joback Method
cpg	690.11	J/molxK	697.17	Joback Method
cpg	706.74	J/molxK	727.15	Joback Method
cpg	722.53	J/molxK	757.14	Joback Method
cpg	737.50	J/molxK	787.12	Joback Method
cpg	751.66	J/molxK	817.10	Joback Method
dvisc	0.0035901	Paxs	325.62	Joback Method

dvisc	0.0012334	Paxs	377.55	Joback Method
dvisc	0.0005486	Paxs	429.48	Joback Method
dvisc	0.0002907	Paxs	481.41	Joback Method
dvisc	0.0001743	Paxs	533.34	Joback Method
dvisc	0.0001144	Paxs	585.27	Joback Method
dvisc	0.0000804	Paxs	637.20	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U378242&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U378242&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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