

4-Cyanobenzoic acid, hexadecyl ester

Inchi:	InChI=1S/C24H37NO2/c1-2-3-4-5-6-7-8-9-10-11-12-13-14-15-20-27-24(26)23-18-16-22(
InchiKey:	WXBONVHCLKVXBG-UHFFFAOYSA-N
Formula:	C24H37NO2
SMILES:	CCCCCCCCCCCCCCCCOC(=O)c1ccc(C#N)cc1
Mol. weight [g/mol]:	371.56

Physical Properties

Property code	Value	Unit	Source
gf	153.24	kJ/mol	Joback Method
hf	-393.55	kJ/mol	Joback Method
hfus	55.86	kJ/mol	Joback Method
hvap	91.59	kJ/mol	Joback Method
log10ws	-8.35		Crippen Method
logp	7.196		Crippen Method
mcvol	334.080	ml/mol	McGowan Method
pc	985.78	kPa	Joback Method
rinqol	2848.00		NIST Webbook
tb	958.55	K	Joback Method
tc	1174.06	K	Joback Method
tf	536.33	K	Joback Method
vc	1.321	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1101.17	J/molxK	958.55	Joback Method
cpg	1117.53	J/molxK	994.47	Joback Method
cpg	1132.68	J/molxK	1030.39	Joback Method
cpg	1146.65	J/molxK	1066.30	Joback Method
cpg	1159.52	J/molxK	1102.22	Joback Method
cpg	1171.33	J/molxK	1138.14	Joback Method
cpg	1182.15	J/molxK	1174.06	Joback Method

Sources

NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=U299840&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci9903071
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mccvol:	McGowan's characteristic volume
pc:	Critical Pressure
rinpol:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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