

# E-5-tetradecenyl acetate

<b>Inchi:</b>	InChI=1S/C16H30O2/c1-3-4-5-6-7-8-9-10-11-12-13-14-15-18-16(2)17/h10-11H,3-9,12-15
<b>InchiKey:</b>	IAGBQBDKOCV GCC-ZHACJKMWSA-N
<b>Formula:</b>	C16H30O2
<b>SMILES:</b>	CCCCCCCC=CCCCOC(C)=O
<b>Mol. weight [g/mol]:</b>	254.41

## Physical Properties

Property code	Value	Unit	Source
gf	-69.86	kJ/mol	Joback Method
hf	-501.15	kJ/mol	Joback Method
hfus	40.18	kJ/mol	Joback Method
hvap	60.32	kJ/mol	Joback Method
log10ws	-5.24		Crippen Method
logp	5.027		Crippen Method
mcvol	239.440	ml/mol	McGowan Method
pc	1417.57	kPa	Joback Method
rinpol	1795.00		NIST Webbook
ripol	2108.00		NIST Webbook
ripol	2108.00		NIST Webbook
tb	645.93	K	Joback Method
tc	818.34	K	Joback Method
tf	337.16	K	Joback Method
vc	0.935	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	655.12	J/molxK	645.93	Joback Method
cpg	734.94	J/molxK	789.61	Joback Method
cpg	720.44	J/molxK	760.87	Joback Method
cpg	705.23	J/molxK	732.14	Joback Method
cpg	689.29	J/molxK	703.40	Joback Method
cpg	672.59	J/molxK	674.67	Joback Method
cpg	748.76	J/molxK	818.34	Joback Method

dvisc	0.0001048	Paxs	645.93	Joback Method
dvisc	0.0001400	Paxs	594.47	Joback Method
dvisc	0.0001976	Paxs	543.01	Joback Method
dvisc	0.0002997	Paxs	491.55	Joback Method
dvisc	0.0005012	Paxs	440.08	Joback Method
dvisc	0.0009605	Paxs	388.62	Joback Method
dvisc	0.0022449	Paxs	337.16	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R86734&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R86734&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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