

# N,N-Dimethyl-N'-pentyl-isobutyramidine

<b>Inchi:</b>	InChI=1S/C11H24N2/c1-6-7-8-9-12-11(10(2)3)13(4)5/h10H,6-9H2,1-5H3
<b>InchiKey:</b>	TTYVDSFXAAKXCM-UHFFFAOYSA-N
<b>Formula:</b>	C11H24N2
<b>SMILES:</b>	CCCCCN=C(C(C)C)N(C)C
<b>Mol. weight [g/mol]:</b>	184.32

## Physical Properties

Property code	Value	Unit	Source
hf	-135.69	kJ/mol	Joback Method
hvap	45.13	kJ/mol	Joback Method
log10ws	-2.41		Crippen Method
logp	2.793		Crippen Method
mcvol	181.510	ml/mol	McGowan Method
pc	1783.35	kPa	Joback Method
rinpol	1230.00		NIST Webbook
rinpol	1230.00		NIST Webbook
tb	539.64	K	Joback Method
tc	723.28	K	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R162753&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R162753&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307I">http://pubs.acs.org/doi/abs/10.1021/ci990307I</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions

<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature

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