

Diethylmalonic acid, 4-cyanophenyl octyl ester

Inchi:	InChI=1S/C22H31NO4/c1-4-7-8-9-10-11-16-26-20(24)22(5-2,6-3)21(25)27-19-14-12-18(
InchiKey:	DSAJTWSPDVTGTD-UHFFFAOYSA-N
Formula:	C22H31NO4
SMILES:	CCCCCCCCOC(=O)C(CC)(CC)C(=O)Oc1ccc(C#N)cc1
Mol. weight [g/mol]:	373.49

Physical Properties

Property code	Value	Unit	Source
gf	-94.68	kJ/mol	Joback Method
hf	-605.82	kJ/mol	Joback Method
hfus	46.05	kJ/mol	Joback Method
hvap	95.00	kJ/mol	Joback Method
log10ws	-6.20		Crippen Method
logp	5.174		Crippen Method
mvol	313.340	ml/mol	McGowan Method
pc	1166.43	kPa	Joback Method
rinpol	2607.00		NIST Webbook
rinpol	2607.00		NIST Webbook
tb	985.85	K	Joback Method
tc	1209.67	K	Joback Method
tf	588.37	K	Joback Method
vc	1.222	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1020.51	J/mol×K	985.85	Joback Method
cpg	1033.93	J/mol×K	1023.15	Joback Method
cpg	1046.14	J/mol×K	1060.46	Joback Method
cpg	1057.19	J/mol×K	1097.76	Joback Method
cpg	1067.13	J/mol×K	1135.06	Joback Method
cpg	1076.04	J/mol×K	1172.37	Joback Method
cpg	1083.96	J/mol×K	1209.67	Joback Method

Sources

McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=U369606&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307I
Crippen Method:	https://www.cheméo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
h vap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
r in pol:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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