

# Dimethylmalonic acid, 2-naphthylmethyl pentyl ester

Inchi:	InChI=1S/C21H26O4/c1-4-5-8-13-24-19(22)21(2,3)20(23)25-15-16-11-12-17-9-6-7-10-18
InchiKey:	CCAJMZZWYCLHR-UHFFFAOYSA-N
Formula:	C21H26O4
SMILES:	CCCCOC(=O)C(C)(C)C(=O)OCc1ccc2ccccc2c1
Mol. weight [g/mol]:	342.43

## Physical Properties

Property code	Value	Unit	Source
gf	-129.63	kJ/mol	Joback Method
hf	-558.99	kJ/mol	Joback Method
hfus	38.98	kJ/mol	Joback Method
hvap	83.93	kJ/mol	Joback Method
log10ws	-5.82		Crippen Method
logp	4.643		Crippen Method
mvol	278.410	ml/mol	McGowan Method
pc	1523.50	kPa	Joback Method
rinpol	2551.00		NIST Webbook
rinpol	2551.00		NIST Webbook
tb	879.87	K	Joback Method
tc	1099.65	K	Joback Method
tf	544.81	K	Joback Method
vc	1.062	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	861.88	J/molxK	879.87	Joback Method
cpg	926.43	J/molxK	1063.02	Joback Method
cpg	915.38	J/molxK	1026.39	Joback Method
cpg	903.48	J/molxK	989.76	Joback Method
cpg	890.65	J/molxK	953.13	Joback Method
cpg	876.81	J/molxK	916.50	Joback Method
cpg	936.71	J/molxK	1099.65	Joback Method
dvisc	0.0000742	Paxs	879.87	Joback Method

dvisc	0.0000931	Paxs	824.03	Joback Method
dvisc	0.0001208	Paxs	768.18	Joback Method
dvisc	0.0001632	Paxs	712.34	Joback Method
dvisc	0.0002321	Paxs	656.50	Joback Method
dvisc	0.0003525	Paxs	600.65	Joback Method
dvisc	0.0005831	Paxs	544.81	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U363858&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U363858&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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