# 2(1H)-Pyridone, 6-methyl-

Other names: 6-Methyl-2-pyridinol

2(1H)-Pyridinone, 6-methyl-

2-Methyl-6-pyridone

6-Methyl-2-hydroxypyridine

6-Methyl-2-pyridinone

6-Methyl-2-pyridone

2-Hydroxy-6-methylpyridine 6-Methyl-2-pyridyl alcohol 6-Methyl-2(1H)-pyridone 2-Methyl-6-hydroxypyridine 6-Methyl-2(1H)-pyridinone

6-methylpyridin-2-ol

InChl=1S/C6H7NO/c1-5-3-2-4-6(8)7-5/h2-4H,1H3,(H,7,8)

InchiKey: JEAVIRYCMBDJIU-UHFFFAOYSA-N

Formula: C6H7NO

SMILES: Cc1cccc(=O)[nH]1

Mol. weight [g/mol]: 109.13 CAS: 3279-76-3

## **Physical Properties**

Property code	Value	Unit	Source
chs	-3149.20 ± 1.90	kJ/mol	NIST Webbook
hf	-120.30 ± 2.50	kJ/mol	NIST Webbook
hfs	-212.30 ± 2.10	kJ/mol	NIST Webbook
hsub	92.00 ± 1.30	kJ/mol	NIST Webbook
hsub	92.00 ± 1.30	kJ/mol	NIST Webbook
ie	$8.19 \pm 0.03$	eV	NIST Webbook
log10ws	-0.59		Crippen Method
logp	0.201		Crippen Method
mcvol	87.490	ml/mol	McGowan Method

### Sources

**Crippen Method:** 

McGowan Method: http://link.springer.com/article/10.1007/BF02311772

NIST Webbook: http://webbook.nist.gov/cgi/cbook.cgi?ID=C3279763&Units=SI

Crippen Method: http://pubs.acs.org/doi/abs/10.1021/ci990307l

### Legend

**chs:** Standard solid enthalpy of combustion

**hf:** Enthalpy of formation at standard conditions

**hfs:** Solid phase enthalpy of formation at standard conditions

**hsub:** Enthalpy of sublimation at standard conditions

ie: Ionization energy

log10ws: Log10 of Water solubility in mol/llogp: Octanol/Water partition coefficientmcvol: McGowan's characteristic volume

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