

N,N-Dimethyl-N'-(3-nitrophenyl)-isobutyramidine

Inchi: InChI=1S/C12H17N3O2/c1-9(2)12(14(3)4)13-10-6-5-7-11(8-10)15(16)17/h5-9H,1-4H3
InchiKey: CGDLVAPBYFTONO-UHFFFAOYSA-N
Formula: C12H17N3O2
SMILES: CC(C)C(=Nc1cccc([N+](=O)[O-])c1)N(C)C
Mol. weight [g/mol]: 235.28

Physical Properties

| Property code | Value | Unit | Source |
|---------------|---------|--------|----------------|
| hf | 57.97 | kJ/mol | Joback Method |
| hvap | 66.88 | kJ/mol | Joback Method |
| log10ws | -3.19 | | Crippen Method |
| logp | 2.842 | | Crippen Method |
| mcvol | 189.260 | ml/mol | McGowan Method |
| pc | 2191.78 | kPa | Joback Method |
| rinpol | 2000.00 | | NIST Webbook |
| tb | 746.02 | K | Joback Method |
| tc | 989.34 | K | Joback Method |

Sources

Crippen Method: <http://pubs.acs.org/doi/abs/10.1021/ci9903071>
Crippen Method: https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method: https://en.wikipedia.org/wiki/Joback_method
McGowan Method: <http://link.springer.com/article/10.1007/BF02311772>
NIST Webbook: <http://webbook.nist.gov/cgi/cbook.cgi?ID=R162451&Units=SI>

Legend

hf: Enthalpy of formation at standard conditions
hvap: Enthalpy of vaporization at standard conditions
log10ws: Log10 of Water solubility in mol/l

| | |
|----------------|-------------------------------------|
| logp: | Octanol/Water partition coefficient |
| mcvol: | McGowan's characteristic volume |
| pc: | Critical Pressure |
| rinpol: | Non-polar retention indices |
| tb: | Normal Boiling Point Temperature |
| tc: | Critical Temperature |

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