

# 1,1'-Biphenyl, 2-nitro-

<b>Other names:</b>	Biphenyl, 2-nitro- o-Nitrobiphenyl o-Nitrodiphenyl Onb 2-Nitrobiphenyl 2-Nitrodiphenyl
<b>Inchi:</b>	InChI=1S/C12H9NO2/c14-13(15)12-9-5-4-8-11(12)10-6-2-1-3-7-10/h1-9H
<b>InchiKey:</b>	YOJKKXRJMXIKSR-UHFFFAOYSA-N
<b>Formula:</b>	C12H9NO2
<b>SMILES:</b>	O=[N+](O)c1ccccc1-c1ccccc1
<b>Mol. weight [g/mol]:</b>	199.21
<b>CAS:</b>	86-00-0

## Physical Properties

Property code	Value	Unit	Source
ea	1.07 ± 0.10	eV	NIST Webbook
gf	300.90	kJ/mol	Joback Method
hf	159.82	kJ/mol	Joback Method
hfus	25.89	kJ/mol	Joback Method
hvap	64.11	kJ/mol	Joback Method
log10ws	-4.73		Crippen Method
logp	3.262		Crippen Method
mcvol	149.840	ml/mol	McGowan Method
pc	3415.86	kPa	Joback Method
rinpol	1669.00		NIST Webbook
rinpol	1670.00		NIST Webbook
rinpol	1698.00		NIST Webbook
rinpol	1700.00		NIST Webbook
rinpol	1700.00		NIST Webbook
rinpol	1669.00		NIST Webbook
rinpol	1669.00		NIST Webbook
rinpol	1686.00		NIST Webbook
rinpol	1696.33		NIST Webbook
rinpol	290.25		NIST Webbook
rinpol	288.70		NIST Webbook
rinpol	288.56		NIST Webbook
rinpol	284.73		NIST Webbook

rinpol	288.54		NIST Webbook
rinpol	287.61		NIST Webbook
rinpol	290.25		NIST Webbook
rinpol	285.80		NIST Webbook
rinpol	285.80		NIST Webbook
rinpol	1669.00		NIST Webbook
rinpol	1686.00		NIST Webbook
rinpol	1691.22		NIST Webbook
tb	593.20	K	NIST Webbook
tc	959.45	K	Joback Method
tf	433.97	K	Joback Method
vc	0.574	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	373.70	J/mol×K	684.14	Joback Method
cpg	387.39	J/mol×K	730.02	Joback Method
cpg	399.76	J/mol×K	775.91	Joback Method
cpg	410.91	J/mol×K	821.79	Joback Method
cpg	420.94	J/mol×K	867.68	Joback Method
cpg	429.96	J/mol×K	913.56	Joback Method
cpg	438.07	J/mol×K	959.45	Joback Method

## Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	440.70	K	1.70	NIST Webbook

## Sources

McGowan Method:

<http://link.springer.com/article/10.1007/BF02311772>

NIST Webbook:

<http://webbook.nist.gov/cgi/cbook.cgi?ID=C86000&Units=SI>

Crippen Method:

<http://pubs.acs.org/doi/abs/10.1021/ci9903071>

Crippen Method:

[https://www.chemeo.com/doc/models/crippen\\_log10ws](https://www.chemeo.com/doc/models/crippen_log10ws)

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>ea:</b>	Electron affinity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpola:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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