

# Benzenemethanol, «alpha»,«alpha»-dimethyl-

<b>Other names:</b>	1-Hydroxycumene 1-Methyl-1-phenylethanol 2-Hydroxy-2-phenylpropane 2-Phenyl-2-propanol 2-Phenylisopropanol 2-Propanol, 2-phenyl- 2-phenylpropan-2-ol Benzenemethanol, .alpha.,.alpha.-dimethyl- Benzyl alcohol, «alpha»,«alpha»-dimethyl- Benzyl alcohol, Â«alphaÂ»,Â«alphaÂ»-dimethyl- Dimethyl phenyl carbinol Dimethylphenylcarbinol Dimethylphenylmethanol NSC 1261 NSC 212537 Phenyldimethylcarbinol «alpha»,«alpha»-Dimethylbenzenemethanol «alpha»,«alpha»-Dimethylbenzyl alcohol «alpha»-Cumyl alcohol Â«alphaÂ»,Â«alphaÂ»-Dimethylbenzenemethanol Â«alphaÂ»,Â«alphaÂ»-Dimethylbenzyl alcohol Â«alphaÂ»-Cumyl alcohol
<b>Inchi:</b>	InChI=1S/C9H12O/c1-9(2,10)8-6-4-3-5-7-8/h3-7,10H,1-2H3
<b>InchiKey:</b>	BDCFWIDZNLCTMF-UHFFFAOYSA-N
<b>Formula:</b>	C9H12O
<b>SMILES:</b>	CC(C)(O)c1ccccc1
<b>Mol. weight [g/mol]:</b>	136.19
<b>CAS:</b>	617-94-7

## Physical Properties

Property code	Value	Unit	Source
chs	-5006.20 ± 2.20	kJ/mol	NIST Webbook
gf	3.33	kJ/mol	Joback Method
hf	-153.54	kJ/mol	Joback Method
hfs	-250.40 ± 2.30	kJ/mol	NIST Webbook
hfus	9.78	kJ/mol	Joback Method

hsub	82.30 ± 0.70		kJ/mol	NIST Webbook
hvap	65.00 ± 0.50		kJ/mol	NIST Webbook
log10ws	-2.13			Crippen Method
logp	1.914			Crippen Method
mcvol	119.780		ml/mol	McGowan Method
pc	3718.02		kPa	Joback Method
rinpol	1098.00			NIST Webbook
rinpol	1089.60			NIST Webbook
rinpol	1089.00			NIST Webbook
rinpol	1090.00			NIST Webbook
rinpol	1090.00			NIST Webbook
rinpol	1080.00			NIST Webbook
rinpol	1057.00			NIST Webbook
rinpol	1076.00			NIST Webbook
rinpol	1095.00			NIST Webbook
rinpol	1076.00			NIST Webbook
rinpol	1098.00			NIST Webbook
rinpol	1057.00			NIST Webbook
rinpol	1098.00			NIST Webbook
rinpol	1102.00			NIST Webbook
ripol	1776.00			NIST Webbook
ripol	1782.00			NIST Webbook
ripol	1770.00			NIST Webbook
ripol	1759.00			NIST Webbook
ripol	1779.00			NIST Webbook
ripol	1770.00			NIST Webbook
ripol	1770.00			NIST Webbook
tb	475.20		K	NIST Webbook
tc	728.04		K	Joback Method
tf	346.00 ± 3.00		K	NIST Webbook
tf	310.00 ± 1.00		K	NIST Webbook
vc	0.440		m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	324.80	J/mol×K	693.52	Joback Method
cpg	333.56	J/mol×K	728.04	Joback Method
cpg	270.14	J/mol×K	520.95	Joback Method
cpg	282.69	J/mol×K	555.46	Joback Method
cpg	294.38	J/mol×K	589.98	Joback Method

cpg	305.26	J/mol×K	624.49	Joback Method
cpg	315.39	J/mol×K	659.01	Joback Method
dvisc	0.0001992	Paxs	480.93	Joback Method
dvisc	0.0001204	Paxs	520.95	Joback Method
dvisc	0.0212186	Paxs	280.85	Joback Method
dvisc	0.0052355	Paxs	320.87	Joback Method
dvisc	0.0017619	Paxs	360.88	Joback Method
dvisc	0.0007369	Paxs	400.90	Joback Method
dvisc	0.0003611	Paxs	440.92	Joback Method
hsubt	87.50 ± 2.60	kJ/mol	294.00	NIST Webbook
hsubt	82.80 ± 0.70	kJ/mol	289.00	NIST Webbook
hvapt	52.90	kJ/mol	407.00	NIST Webbook
hvapt	63.40 ± 0.50	kJ/mol	324.50	NIST Webbook

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.64941e+01
Coeff. B	-5.17060e+03
Coeff. C	-3.97590e+01
Temperature range (K), min.	358.80
Temperature range (K), max.	502.14

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/T + C \cdot \ln(T) + D \cdot T^2$
Coeff. A	1.07927e+02
Coeff. B	-1.09574e+04
Coeff. C	-1.32851e+01
Coeff. D	7.56204e-06
Temperature range (K), min.	379.15
Temperature range (K), max.	431.15

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C617947&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C617947&amp;Units=SI</a>
<b>The Yaws Handbook of Vapor Pressure: KDB Vapor Pressure Data:</b>	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a> <a href="https://www.cheric.org/research/kdb/hcprop/showprop.php?cmpid=882">https://www.cheric.org/research/kdb/hcprop/showprop.php?cmpid=882</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>KDB:</b>	<a href="https://www.cheric.org/files/research/kdb/mol/mol882.mol">https://www.cheric.org/files/research/kdb/mol/mol882.mol</a>

## Legend

<b>chs:</b>	Standard solid enthalpy of combustion
<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfs:</b>	Solid phase enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hsub:</b>	Enthalpy of sublimation at standard conditions
<b>hsubt:</b>	Enthalpy of sublimation at a given temperature
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>hvapt:</b>	Enthalpy of vaporization at a given temperature
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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