

# Glutaric acid, hexyl 3-phenylprop-2-enyl ester

<b>Inchi:</b>	InChI=1S/C20H28O4/c1-2-3-4-8-16-23-19(21)14-9-15-20(22)24-17-10-13-18-11-6-5-7-12
<b>InchiKey:</b>	BZQLSNPIGQYAQL-JLHYYAGUSA-N
<b>Formula:</b>	C20H28O4
<b>SMILES:</b>	CCCCCOC(=O)CCCC(=O)OCC=Cc1ccccc1
<b>Mol. weight [g/mol]:</b>	332.43

## Physical Properties

Property code	Value	Unit	Source
gf	-157.69	kJ/mol	Joback Method
hf	-591.98	kJ/mol	Joback Method
hfus	47.37	kJ/mol	Joback Method
hvap	80.66	kJ/mol	Joback Method
log10ws	-5.04		Crippen Method
logp	4.537		Crippen Method
mcvol	279.480	ml/mol	McGowan Method
pc	1410.13	kPa	Joback Method
rinpola	2615.00		NIST Webbook
tb	840.42	K	Joback Method
tc	1043.70	K	Joback Method
tf	480.82	K	Joback Method
vc	1.075	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	852.00	J/molxK	840.42	Joback Method
cpg	920.37	J/molxK	1009.82	Joback Method
cpg	908.69	J/molxK	975.94	Joback Method
cpg	896.05	J/molxK	942.06	Joback Method
cpg	882.42	J/molxK	908.18	Joback Method
cpg	867.75	J/molxK	874.30	Joback Method
cpg	931.14	J/molxK	1043.70	Joback Method
dvisc	0.0000472	Paxs	840.42	Joback Method
dvisc	0.0000617	Paxs	780.49	Joback Method

dvisc	0.0000844	Paxs	720.55	Joback Method
dvisc	0.0001223	Paxs	660.62	Joback Method
dvisc	0.0001906	Paxs	600.69	Joback Method
dvisc	0.0003279	Paxs	540.75	Joback Method
dvisc	0.0006456	Paxs	480.82	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U359891&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U359891&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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