

# Tetraethyl sulphamide

<b>Other names:</b>	N,N,N',N'-Tetraethylsulfamide N-(diethylsulfamoyl)-N-ethylethanamine
<b>Inchi:</b>	InChI=1S/C8H20N2O2S/c1-5-9(6-2)13(11,12)10(7-3)8-4/h5-8H2,1-4H3
<b>InchiKey:</b>	IYIAWAACGTUPCC-UHFFFAOYSA-N
<b>Formula:</b>	C8H20N2O2S
<b>SMILES:</b>	CCN(CC)S(=O)(=O)N(CC)CC
<b>Mol. weight [g/mol]:</b>	208.32
<b>CAS:</b>	2832-49-7

## Physical Properties

Property code	Value	Unit	Source
gf	-230.50	kJ/mol	Joback Method
hf	-526.74	kJ/mol	Joback Method
hfl	-635.50 ± 9.90	kJ/mol	NIST Webbook
hfus	33.90	kJ/mol	Joback Method
hvap	56.12	kJ/mol	Joback Method
log10ws	-1.14		Aqueous Solubility Prediction Method
logp	0.915		Crippen Method
mcvol	171.630	ml/mol	McGowan Method
pc	2969.80	kPa	Joback Method
tb	455.10	K	Joback Method
tc	614.57	K	Joback Method
tf	283.42	K	Joback Method
vc	0.645	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	444.10	J/mol×K	587.99	Joback Method
cpg	373.92	J/mol×K	455.10	Joback Method
cpg	389.16	J/mol×K	481.68	Joback Method
cpg	403.79	J/mol×K	508.26	Joback Method
cpg	417.81	J/mol×K	534.83	Joback Method

cpg	431.25	J/mol×K	561.41	Joback Method
cpg	456.39	J/mol×K	614.57	Joback Method
hvapt	59.10	kJ/mol	467.50	NIST Webbook

## Sources

**Aqueous Solubility Prediction Method:** <http://onschallenge.wikispaces.com/file/view/AqueousDataset002.xlsx/351826032/AqueousDa>

**McGowan Method:** <http://link.springer.com/article/10.1007/BF02311772>

**NIST Webbook:** <http://webbook.nist.gov/cgi/cbook.cgi?ID=C2832497&Units=SI>

**Crippen Method:** <http://pubs.acs.org/doi/abs/10.1021/ci9903071>

**Joback Method:** [https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfl:</b>	Liquid phase enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>hvapt:</b>	Enthalpy of vaporization at a given temperature
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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