

# Bicyclo[2.2.1]hept-5-ene-2-carboxylic acid, methyl ester, endo-

<b>Other names:</b>	Methyl bicyclo[2.2.1]hept-5-ene-2-carboxylate, endo-endo-Bicyclo[2.2.1]hept-5-en-2-carboxylic acid, methyl ester
<b>Inchi:</b>	InChI=1S/C9H12O2/c1-11-9(10)8-5-6-2-3-7(8)4-6/h2-3,6-8H,4-5H2,1H3/t6-,7-,8-/m1/s1
<b>InchiKey:</b>	RMAZRAQKPTXZNL-BWZBUEFSSA-N
<b>Formula:</b>	C9H12O2
<b>SMILES:</b>	COC(=O)C1CC2C=CC1C2
<b>Mol. weight [g/mol]:</b>	152.19
<b>CAS:</b>	2903-75-5

## Physical Properties

Property code	Value	Unit	Source
gf	-77.37	kJ/mol	Joback Method
hf	-297.01	kJ/mol	Joback Method
hfus	18.32	kJ/mol	Joback Method
hvap	44.77	kJ/mol	Joback Method
log10ws	-1.37		Crippen Method
logp	1.372		Crippen Method
mvol	119.090	ml/mol	McGowan Method
pc	3272.78	kPa	Joback Method
rinpol	1103.00		NIST Webbook
ripol	1527.00		NIST Webbook
tb	493.85	K	Joback Method
tc	704.70	K	Joback Method
tf	292.23	K	Joback Method
vc	0.455	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	278.17	J/molxK	493.85	Joback Method
cpg	293.60	J/molxK	528.99	Joback Method
cpg	308.08	J/molxK	564.13	Joback Method
cpg	321.67	J/molxK	599.27	Joback Method
cpg	334.41	J/molxK	634.41	Joback Method

cpg	346.34	J/molxK	669.56	Joback Method
cpg	357.51	J/molxK	704.70	Joback Method
dvisc	0.0013177	Paxs	292.23	Joback Method
dvisc	0.0011587	Paxs	325.83	Joback Method
dvisc	0.0010436	Paxs	359.44	Joback Method
dvisc	0.0009570	Paxs	393.04	Joback Method
dvisc	0.0008896	Paxs	426.64	Joback Method
dvisc	0.0008358	Paxs	460.25	Joback Method
dvisc	0.0007920	Paxs	493.85	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C2903755&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C2903755&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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