

# 1,2,6-trimethoxy-4-propylbenzene

<b>Inchi:</b>	InChI=1S/C12H18O3/c1-5-6-9-7-10(13-2)12(15-4)11(8-9)14-3/h7-8H,5-6H2,1-4H3
<b>InchiKey:</b>	AOVUTAOJQZOAGA-UHFFFAOYSA-N
<b>Formula:</b>	C12H18O3
<b>SMILES:</b>	CCc1cc(OC)c(OC)c(OC)c1
<b>Mol. weight [g/mol]:</b>	210.27

## Physical Properties

Property code	Value	Unit	Source
gf	-181.32	kJ/mol	Joback Method
hf	-485.55	kJ/mol	Joback Method
hfus	23.27	kJ/mol	Joback Method
hvap	53.80	kJ/mol	Joback Method
log10ws	-3.05		Crippen Method
logp	2.665		Crippen Method
mcvol	173.790	ml/mol	McGowan Method
pc	2193.84	kPa	Joback Method
rinpol	1730.00		NIST Webbook
rinpol	1730.00		NIST Webbook
tb	582.84	K	Joback Method
tc	780.29	K	Joback Method
tf	355.67	K	Joback Method
vc	0.653	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	424.58	J/molxK	582.84	Joback Method
cpg	439.63	J/molxK	615.75	Joback Method
cpg	454.08	J/molxK	648.66	Joback Method
cpg	467.91	J/molxK	681.57	Joback Method
cpg	481.11	J/molxK	714.47	Joback Method
cpg	493.65	J/molxK	747.38	Joback Method
cpg	505.52	J/molxK	780.29	Joback Method
dvisc	0.0006752	Paxs	355.67	Joback Method

dvisc	0.0004279	Paxs	393.53	Joback Method
dvisc	0.0002938	Paxs	431.39	Joback Method
dvisc	0.0002143	Paxs	469.25	Joback Method
dvisc	0.0001639	Paxs	507.12	Joback Method
dvisc	0.0001301	Paxs	544.98	Joback Method
dvisc	0.0001064	Paxs	582.84	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R322222&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R322222&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>g<sub>f</sub>:</b>	Standard Gibbs free energy of formation
<b>h<sub>f</sub>:</b>	Enthalpy of formation at standard conditions
<b>h<sub>fus</sub>:</b>	Enthalpy of fusion at standard conditions
<b>h<sub>vap</sub>:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>log<sub>p</sub>:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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