

Benzoic acid, 2,4,5-trichlorophenyl ester

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|-----------------------------|---|
| Inchi: | InChI=1S/C13H7Cl3O2/c14-9-6-11(16)12(7-10(9)15)18-13(17)8-4-2-1-3-5-8/h1-7H |
| InchiKey: | MDWSVFNVPWZKMU-UHFFFAOYSA-N |
| Formula: | C13H7Cl3O2 |
| SMILES: | O=C(Oc1cc(Cl)c(Cl)cc1Cl)c1ccccc1 |
| Mol. weight [g/mol]: | 301.55 |

Physical Properties

| Property code | Value | Unit | Source |
|---------------|---------|---------|----------------|
| gf | -15.20 | kJ/mol | Joback Method |
| hf | -165.02 | kJ/mol | Joback Method |
| hfus | 31.72 | kJ/mol | Joback Method |
| hvap | 73.38 | kJ/mol | Joback Method |
| log10ws | -5.61 | | Crippen Method |
| logp | 4.866 | | Crippen Method |
| mcvol | 190.670 | ml/mol | McGowan Method |
| pc | 2729.71 | kPa | Joback Method |
| rinpol | 2195.00 | | NIST Webbook |
| rinpol | 2195.00 | | NIST Webbook |
| tb | 753.72 | K | Joback Method |
| tc | 1011.22 | K | Joback Method |
| tf | 488.59 | K | Joback Method |
| vc | 0.719 | m3/kmol | Joback Method |

Temperature Dependent Properties

| Property code | Value | Unit | Temperature [K] | Source |
|---------------|-----------|---------|-----------------|---------------|
| cpg | 429.57 | J/molxK | 753.72 | Joback Method |
| cpg | 471.87 | J/molxK | 968.30 | Joback Method |
| cpg | 465.30 | J/molxK | 925.39 | Joback Method |
| cpg | 457.83 | J/molxK | 882.47 | Joback Method |
| cpg | 449.40 | J/molxK | 839.55 | Joback Method |
| cpg | 440.00 | J/molxK | 796.64 | Joback Method |
| cpg | 477.58 | J/molxK | 1011.22 | Joback Method |
| dvisc | 0.0001318 | Paxs | 753.72 | Joback Method |

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|-------|-----------|------|--------|---------------|
| dvisc | 0.0001598 | Paxs | 709.53 | Joback Method |
| dvisc | 0.0001988 | Paxs | 665.34 | Joback Method |
| dvisc | 0.0002550 | Paxs | 621.15 | Joback Method |
| dvisc | 0.0003399 | Paxs | 576.97 | Joback Method |
| dvisc | 0.0004751 | Paxs | 532.78 | Joback Method |
| dvisc | 0.0007055 | Paxs | 488.59 | Joback Method |

Sources

| | |
|------------------------|---|
| Joback Method: | https://en.wikipedia.org/wiki/Joback_method |
| McGowan Method: | http://link.springer.com/article/10.1007/BF02311772 |
| NIST Webbook: | http://webbook.nist.gov/cgi/cbook.cgi?ID=U360686&Units=SI |
| Crippen Method: | http://pubs.acs.org/doi/abs/10.1021/ci990307l |
| Crippen Method: | https://www.chemeo.com/doc/models/crippen_log10ws |

Legend

| | |
|-----------------|---|
| cpg: | Ideal gas heat capacity |
| dvisc: | Dynamic viscosity |
| gf: | Standard Gibbs free energy of formation |
| hf: | Enthalpy of formation at standard conditions |
| hfus: | Enthalpy of fusion at standard conditions |
| hvap: | Enthalpy of vaporization at standard conditions |
| log10ws: | Log10 of Water solubility in mol/l |
| logp: | Octanol/Water partition coefficient |
| mcvol: | McGowan's characteristic volume |
| pc: | Critical Pressure |
| rinpol: | Non-polar retention indices |
| tb: | Normal Boiling Point Temperature |
| tc: | Critical Temperature |
| tf: | Normal melting (fusion) point |
| vc: | Critical Volume |

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