

# 3-Bromo-4-methoxybenzaldehyde

<b>Other names:</b>	Benzaldehyde, 3-bromo-4-methoxy- p-Anisaldehyde, 3-bromo- 3-Bromo-p-anisaldehyde
<b>Inchi:</b>	InChI=1S/C8H7BrO2/c1-11-8-3-2-6(5-10)4-7(8)9/h2-5H,1H3
<b>InchiKey:</b>	QMPNFQLVIGPNEI-UHFFFAOYSA-N
<b>Formula:</b>	C8H7BrO2
<b>SMILES:</b>	<chem>COc1ccc(C=O)cc1Br</chem>
<b>Mol. weight [g/mol]:</b>	215.04
<b>CAS:</b>	34841-06-0

## Physical Properties

Property code	Value	Unit	Source
gf	-80.57	kJ/mol	Joback Method
hf	-186.33	kJ/mol	Joback Method
hfus	18.50	kJ/mol	Joback Method
hvap	52.57	kJ/mol	Joback Method
log10ws	-3.00		Crippen Method
logp	2.270		Crippen Method
mcvol	124.760	ml/mol	McGowan Method
pc	4114.41	kPa	Joback Method
tb	556.32	K	Joback Method
tc	788.01	K	Joback Method
tf	355.41	K	Joback Method
vc	0.472	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	247.43	J/molxK	556.32	Joback Method
cpg	290.09	J/molxK	749.39	Joback Method
cpg	282.68	J/molxK	710.78	Joback Method
cpg	274.72	J/molxK	672.16	Joback Method
cpg	266.20	J/molxK	633.55	Joback Method
cpg	257.11	J/molxK	594.93	Joback Method

cpg	296.97	J/mol×K	788.01	Joback Method
dvisc	0.0002769	Paxs	556.32	Joback Method
dvisc	0.0003327	Paxs	522.83	Joback Method
dvisc	0.0004099	Paxs	489.35	Joback Method
dvisc	0.0005208	Paxs	455.87	Joback Method
dvisc	0.0006871	Paxs	422.38	Joback Method
dvisc	0.0009510	Paxs	388.89	Joback Method
dvisc	0.0013992	Paxs	355.41	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C34841060&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C34841060&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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