3-Hexenal, (Z)-

Other names:	(3Z)-3-Hexenal
	(Z)-3-Hexen-1-al
	(Z)-3-Hexenal
	(Z)-Hex-3enal
	3-(Z)-Hexenal
	Hex-cis-3-enal
	cis-3-Hexenal
	cis-«beta»,«gamma»-Hexylenic aldehyde
	cis-Â«betaÂ»,Â«gammaÂ»-Hexylenic aldehyde
Inchi:	InChI=1S/C6H10O/c1-2-3-4-5-6-7/h3-4,6H,2,5H2,1H3/b4-3-
InchiKey:	GXANMBISFKBPEX-ARJAWSKDSA-N
Formula:	C6H10O
SMILES:	CCC=CCC=O
Mol. weight [g/mol]:	98.14
CAS:	6789-80-6

Physical Properties

Property code	Value	Unit	Source
gf	-19.66	kJ/mol	Joback Method
hf	-135.53	kJ/mol	Joback Method
hfus	13.79	kJ/mol	Joback Method
hvap	35.63	kJ/mol	Joback Method
log10ws	-1.47		Crippen Method
logp	1.542		Crippen Method
mcvol	92.670	ml/mol	McGowan Method
pc	3594.25	kPa	Joback Method
rinpol	770.00		NIST Webbook
rinpol	804.00		NIST Webbook
rinpol	800.00		NIST Webbook
rinpol	784.00		NIST Webbook
rinpol	795.00		NIST Webbook
rinpol	795.00		NIST Webbook
rinpol	801.00		NIST Webbook
rinpol	778.00		NIST Webbook
rinpol	800.00		NIST Webbook
rinpol	803.00		NIST Webbook
rinpol	799.00		NIST Webbook

rinpol	803.00	NIST Webbook
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rinpol	796.00	NIST Webbook
rinpol	797.00	NIST Webbook
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ripol	1151.00		NIST Webbook
ripol	1142.00		NIST Webbook
ripol	1144.00		NIST Webbook
ripol	1153.00		NIST Webbook
ripol	1135.00		NIST Webbook
tb	389.50	K	Joback Method
tc	570.95	K	Joback Method
tf	194.30	K	Joback Method
vc	0.368	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	164.69	J/mol×K	389.50	Joback Method
cpg	174.16	J/mol×K	419.74	Joback Method
cpg	183.18	J/mol×K	449.98	Joback Method
cpg	191.76	J/mol×K	480.23	Joback Method
cpg	199.93	J/mol×K	510.47	Joback Method
cpg	207.70	J/mol×K	540.71	Joback Method
cpg	215.09	J/mol×K	570.95	Joback Method
dvisc	0.0038680	Pa×s	194.30	Joback Method
dvisc	0.0017894	Pa×s	226.83	Joback Method
dvisc	0.0010044	Pa×s	259.37	Joback Method
dvisc	0.0006413	Pa×s	291.90	Joback Method
dvisc	0.0004480	Pa×s	324.43	Joback Method
dvisc	0.0003341	Pa×s	356.97	Joback Method
dvisc	0.0002617	Pa×s	389.50	Joback Method

Correlations

Information	Value
Property code	pvap
Equation	ln(Pvp) = A + B/(T + C)
Coeff. A	1.37556e+01
Coeff. B	-3.25383e+03
Coeff. C	-5.07240e+01
Temperature range (K), min.	292.32
Temperature range (K), max.	436.06

Sources

Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=C6789806&Units=SI
The Yaws Handbook of Vapor Pressure:	https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l

Legend

cpg:	Ideal gas heat capacity
dvisc:	Dynamic viscosity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
pvap:	Vapor pressure
rinpol:	Non-polar retention indices
ripol:	Polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume