

# Testosterone cypionate

## Other names:

Androst-4-en-3-one, 17-(3-cyclopentyl-1-oxopropoxy)-, (17«beta»)-  
Andro-Cyp  
Dep-Test  
Depo-testosterone  
Depo-Testosterone cypionate  
Depovirin  
Durandro  
Jectatest  
Malogen CYP  
Pertestis  
T-Ionate-P.A  
Testodrin prolongatum  
Testosterone cyclopentanepropionate  
Testosterone cyclopentylpropionate  
Testosterone 17«beta»-cyclopentanepropionate  
Testosterone 17«beta»-cyclopentylpropionate  
Testosterone 17«beta»-cypionate  
depAndro 100  
depAndro 200  
17-(Cyclopentyl-1-oxopropoxy)androst-4-en-3-one, (17«beta»)-  
Depo  
Depotest  
Virilon  
17«beta»-Hydroxyandrost-4-en-3-one cyclopentylpropionate  
NSC 9157

## Inchi:

InChI=1S/C27H40O3/c1-26-15-13-20(28)17-19(26)8-9-21-22-10-11-24(27(22,2)16-14-23

## InchiKey:

HPFVBGJFAYZEBE-KAOZTRPUSA-N

## Formula:

C27H40O3

## SMILES:

CC12CCC(=O)C=C1CCC1C2CCC2(C)C(OC(=O)CCC3CCCC3)CCC12

## Mol. weight [g/mol]:

412.60

## CAS:

58-20-8

## Physical Properties

Property code	Value	Unit	Source
gf	32.93	kJ/mol	Joback Method
hf	-626.12	kJ/mol	Joback Method

hfus	34.34		kJ/mol	Joback Method
hvap	87.90		kJ/mol	Joback Method
log10ws	-7.26			Crippen Method
logp	6.401			Crippen Method
mcvol	341.700		ml/mol	McGowan Method
pc	1223.41		kPa	Joback Method
tb	1020.14		K	Joback Method
tc	1271.89		K	Joback Method
tf	373.00 ± 1.00		K	NIST Webbook
vc	1.288		m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1339.58	J/mol×K	1020.14	Joback Method
cpg	1373.10	J/mol×K	1062.10	Joback Method
cpg	1407.57	J/mol×K	1104.06	Joback Method
cpg	1443.41	J/mol×K	1146.01	Joback Method
cpg	1481.04	J/mol×K	1187.97	Joback Method
cpg	1520.87	J/mol×K	1229.93	Joback Method
cpg	1563.32	J/mol×K	1271.89	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C58208&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C58208&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions

<b>h<sub>vap</sub>:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>w<sub>s</sub>:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>log<sub>p</sub>:</b>	Octanol/Water partition coefficient
<b>mc<sub>vol</sub>:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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