

# 3-Methylglutakonic acid dimethyl ester

<b>Other names:</b>	3-Methyl-pent-2-enedioic acid dimethyl ester, E trans-2-Pentenedioate, 3-methyl, dimethyl ester
<b>Inchi:</b>	InChI=1S/C8H12O4/c1-6(4-7(9)11-2)5-8(10)12-3/h4H,5H2,1-3H3/b6-4+
<b>InchiKey:</b>	GBQBEQOEGMZCOH-GQCTYLIASA-N
<b>Formula:</b>	C8H12O4
<b>SMILES:</b>	COC(=O)C=C(C)CC(=O)OC
<b>Mol. weight [g/mol]:</b>	172.18

## Physical Properties

Property code	Value	Unit	Source
gf	-379.69	kJ/mol	Joback Method
hf	-590.62	kJ/mol	Joback Method
hfus	20.94	kJ/mol	Joback Method
hvap	51.75	kJ/mol	Joback Method
log10ws	-0.75		Crippen Method
logp	0.669		Crippen Method
mcvol	134.160	ml/mol	McGowan Method
pc	2963.34	kPa	Joback Method
rinpol	1195.00		NIST Webbook
rinpol	1198.00		NIST Webbook
tb	539.06	K	Joback Method
tc	734.03	K	Joback Method
tf	305.20	K	Joback Method
vc	0.512	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	303.31	J/molxK	539.06	Joback Method
cpg	314.42	J/molxK	571.56	Joback Method
cpg	325.05	J/molxK	604.05	Joback Method
cpg	335.20	J/molxK	636.55	Joback Method
cpg	344.87	J/molxK	669.04	Joback Method
cpg	354.06	J/molxK	701.54	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U222064&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U222064&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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