

# (S)-(+)-2-Amino-3-methyl-1-butanol

<b>Other names:</b>	(S)-2-Amino-3-methylbutanol 1-Butanol, 2-amino-3-methyl-, (S)- L-2-amino-3-methylbutan-1-ol L-Valinol
<b>Inchi:</b>	InChI=1S/C5H13NO/c1-4(2)5(6)3-7/h4-5,7H,3,6H2,1-2H3/t5-/m0/s1
<b>InchiKey:</b>	NWYYWIJOWOLJNR-YFKPBYRVSA-N
<b>Formula:</b>	C5H13NO
<b>SMILES:</b>	CC(C)C(N)CO
<b>Mol. weight [g/mol]:</b>	103.16
<b>CAS:</b>	2026-48-4

## Physical Properties

Property code	Value	Unit	Source
gf	-84.03	kJ/mol	Joback Method
hf	-275.53	kJ/mol	Joback Method
hfus	10.94	kJ/mol	Joback Method
hvap	53.27	kJ/mol	Joback Method
log10ws	-0.48		Crippen Method
logp	-0.038		Crippen Method
mcvol	97.160	ml/mol	McGowan Method
pc	4255.16	kPa	Joback Method
tb	477.63	K	Joback Method
tc	659.47	K	Joback Method
tf	260.19	K	Joback Method
vc	0.351	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	219.41	J/molxK	477.63	Joback Method
cpg	228.88	J/molxK	507.94	Joback Method
cpg	237.94	J/molxK	538.24	Joback Method
cpg	246.60	J/molxK	568.55	Joback Method
cpg	254.87	J/molxK	598.85	Joback Method

cpg	262.75	J/mol×K	629.16	Joback Method
cpg	270.28	J/mol×K	659.47	Joback Method

## Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	354.20	K	1.00	NIST Webbook

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.34226e+01
Coeff. B	-3.56060e+03
Coeff. C	-6.21630e+01
Temperature range (K), min.	333.24
Temperature range (K), max.	501.14

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C2026484&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C2026484&amp;Units=SI</a>
<b>The Yaws Handbook of Vapor Pressure:</b>	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation

<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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