

# Phenicarbazide

**Other names:** Hydrazinecarboxamide, 2-phenyl-  
Semicarbazide, 1-phenyl-  
Carbaphen  
Cryogenine  
Cryogenine (pharmaceutical)  
Febrimin  
Phenicarbazid  
Phenygenine  
Phenylsemicarbazide  
1-Carbamoyl-2-phenylhydrazine  
1-Carbamyl-2-phenylhydrazine  
1-Phenylsemicarbazide  
Carbamic acid, 2-phenylhydrazide  
Hydrazine, 1-carbamoyl-2-phenyl-  
Kryogenin  
1-Phenylhydrazinecarboxamide  
2-Phenylhydrazide carbamic acid  
2-Phenylhydrazinecarboxamide  
Fenylsemikarbazid  
NSC 2763

**Inchi:** InChI=1S/C7H9N3O/c8-7(11)10-9-6-4-2-1-3-5-6/h1-5,9H,(H3,8,10,11)

**InchiKey:** AVKHCKXGKPAGEI-UHFFFAOYSA-N

**Formula:** C7H9N3O

**SMILES:** NC(=O)NNc1ccccc1

**Mol. weight [g/mol]:** 151.17

**CAS:** 103-03-7

## Physical Properties

Property code	Value	Unit	Source
gf	236.78	kJ/mol	Joback Method
hf	76.87	kJ/mol	Joback Method
hfus	24.92	kJ/mol	Joback Method
hvap	63.71	kJ/mol	Joback Method
log10ws	-1.85		Crippen Method
logp	0.682		Crippen Method
mcvol	117.240	ml/mol	McGowan Method
pc	5008.59	kPa	Joback Method

tb	612.98	K	Joback Method
tc	847.56	K	Joback Method
tf	433.58	K	Joback Method
vc	0.424	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	285.95	J/mol×K	612.98	Joback Method
cpg	296.72	J/mol×K	652.08	Joback Method
cpg	306.66	J/mol×K	691.17	Joback Method
cpg	315.80	J/mol×K	730.27	Joback Method
cpg	324.19	J/mol×K	769.36	Joback Method
cpg	331.88	J/mol×K	808.46	Joback Method
cpg	338.89	J/mol×K	847.56	Joback Method

## Sources

**Crippen Method:** [https://www.chemeo.com/doc/models/crippen\\_log10ws](https://www.chemeo.com/doc/models/crippen_log10ws)

**Joback Method:** [https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)

**McGowan Method:** <http://link.springer.com/article/10.1007/BF02311772>

**NIST Webbook:** <http://webbook.nist.gov/cgi/cbook.cgi?ID=C103037&Units=SI>

**Crippen Method:** <http://pubs.acs.org/doi/abs/10.1021/ci990307l>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure

<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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