

# Cyclohexanol, 2-methyl-5-(1-methylethenyl)-, (1«alpha»,2«alpha»,5«beta»)-

<b>Other names:</b>	p-Menth-8-en-2-ol, cis-1,2,trans-1,4-Neodihydrocarveol 5-Isopropenyl-2-methylcyclohexanol, (1«alpha»,2«alpha»,5«beta»)-Dihydro carveol neo (1R,2S,5S)-neodihydrocarveol Cyclohexanol, 2-methyl-5-(1-methylethenyl)-, (1R,2S,5S)-rel-Neocarveol, dihydro-neo-iso-Dihydrocarveol
<b>Inchi:</b>	InChI=1S/C10H18O/c1-7(2)9-5-4-8(3)10(11)6-9/h8-11H,1,4-6H2,2-3H3
<b>InchiKey:</b>	KRCZYMFUWVJCLI-UHFFFAOYSA-N
<b>Formula:</b>	C10H18O
<b>SMILES:</b>	C=C(C)C1CCC(C)C(O)C1
<b>Mol. weight [g/mol]:</b>	154.25
<b>CAS:</b>	18675-33-7

## Physical Properties

Property code	Value	Unit	Source
gf	-15.18	kJ/mol	Joback Method
hf	-272.68	kJ/mol	Joback Method
hfus	17.13	kJ/mol	Joback Method
hvap	53.75	kJ/mol	Joback Method
log10ws	-2.65		Crippen Method
logp	2.360		Crippen Method
mcvol	142.470	ml/mol	McGowan Method
pc	2770.08	kPa	Joback Method
rinpol	1189.00		NIST Webbook
rinpol	1212.00		NIST Webbook
rinpol	1210.00		NIST Webbook
rinpol	1226.00		NIST Webbook
rinpol	1232.00		NIST Webbook
rinpol	1180.00		NIST Webbook
rinpol	1192.00		NIST Webbook
rinpol	1210.00		NIST Webbook
rinpol	1220.00		NIST Webbook
rinpol	1189.00		NIST Webbook
ripol	1732.00		NIST Webbook
ripol	1735.00		NIST Webbook

ripol	1783.00		NIST Webbook
ripol	1710.00		NIST Webbook
ripol	1777.00		NIST Webbook
tb	527.15	K	Joback Method
tc	719.80	K	Joback Method
tf	246.46	K	Joback Method
vc	0.527	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	350.71	J/mol×K	527.15	Joback Method
cpg	367.51	J/mol×K	559.26	Joback Method
cpg	383.50	J/mol×K	591.37	Joback Method
cpg	398.69	J/mol×K	623.47	Joback Method
cpg	413.11	J/mol×K	655.58	Joback Method
cpg	426.76	J/mol×K	687.69	Joback Method
cpg	439.67	J/mol×K	719.80	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C18675337&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C18675337&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l

<b>logp:</b>	Octanol/Water partition coefficient
<b>mcpol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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