

# (E)-3-(2-Methoxyphenyl)-2-propenoic acid

<b>Other names:</b>	2-Propenoic acid, 3-(2-methoxyphenyl)-, (E)- (E)-o-methoxycinnamic acid
<b>Inchi:</b>	InChI=1S/C10H10O3/c1-13-9-5-3-2-4-8(9)6-7-10(11)12/h2-7H,1H3,(H,11,12)/b7-6+
<b>InchiKey:</b>	FEGVSPGUHMGGBO-VOTSOKGWSA-N
<b>Formula:</b>	C10H10O3
<b>SMILES:</b>	COc1ccccc1C=CC(=O)O
<b>Mol. weight [g/mol]:</b>	178.18
<b>CAS:</b>	1011-54-7

## Physical Properties

Property code	Value	Unit	Source
gf	-154.42	kJ/mol	Joback Method
hf	-304.48	kJ/mol	Joback Method
hfus	22.39	kJ/mol	Joback Method
hsub	128.80 ± 0.60	kJ/mol	NIST Webbook
hvap	66.59	kJ/mol	Joback Method
ie	8.50	eV	NIST Webbook
log10ws	-1.93		Crippen Method
logp	1.793		Crippen Method
mvol	137.010	ml/mol	McGowan Method
pc	3577.07	kPa	Joback Method
tb	632.49	K	Joback Method
tc	839.79	K	Joback Method
tf	369.30	K	Joback Method
vc	0.510	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	328.56	J/mol×K	632.49	Joback Method
cpg	338.85	J/mol×K	667.04	Joback Method
cpg	348.51	J/mol×K	701.59	Joback Method
cpg	357.56	J/mol×K	736.14	Joback Method
cpg	366.02	J/mol×K	770.69	Joback Method

cpg	373.93	J/molxK	805.24	Joback Method
cpg	381.31	J/molxK	839.79	Joback Method
dvisc	0.0023160	Paxs	369.30	Joback Method
dvisc	0.0008973	Paxs	413.17	Joback Method
dvisc	0.0004170	Paxs	457.03	Joback Method
dvisc	0.0002217	Paxs	500.89	Joback Method
dvisc	0.0001304	Paxs	544.76	Joback Method
dvisc	0.0000831	Paxs	588.62	Joback Method
dvisc	0.0000563	Paxs	632.49	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C1011547&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C1011547&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hsub:</b>	Enthalpy of sublimation at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>ie:</b>	Ionization energy
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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