

# Benzene, 1,2,4,5-tetrafluoro-3,6-bis(trifluoromethyl)-

Other names:	Decafluoro-p-xylol
Inchi:	InChI=1S/C8F10/c9-3-1(7(13,14)15)4(10)6(12)2(5(3)11)8(16,17)18
InchiKey:	WWZNHODHBHVRID-UHFFFAOYSA-N
Formula:	C8F10
SMILES:	Fc1c(F)c(C(F)(F)F)c(F)c(F)c1C(F)(F)F
Mol. weight [g/mol]:	286.07
CAS:	651-89-8

## Physical Properties

Property code	Value	Unit	Source
gf	-1861.68	kJ/mol	Joback Method
hf	-2007.87	kJ/mol	Joback Method
hfus	24.54	kJ/mol	Joback Method
hvap	28.23	kJ/mol	Joback Method
ie	9.90	eV	NIST Webbook
log10ws	-5.01		Crippen Method
logp	4.281		Crippen Method
mcvol	117.520	ml/mol	McGowan Method
pc	2179.52	kPa	Joback Method
rinpol	636.10		NIST Webbook
rinpol	636.10		NIST Webbook
tb	420.26	K	Joback Method
tc	565.89	K	Joback Method
tf	279.68	K	Joback Method
vc	0.533	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	268.40	J/molxK	420.26	Joback Method
cpg	276.38	J/molxK	444.53	Joback Method
cpg	283.93	J/molxK	468.80	Joback Method
cpg	291.06	J/molxK	493.07	Joback Method
cpg	297.79	J/molxK	517.34	Joback Method

cpg	304.13	J/mol×K	541.61	Joback Method
cpg	310.09	J/mol×K	565.89	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C651898&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C651898&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvac:</b>	Enthalpy of vaporization at standard conditions
<b>ie:</b>	Ionization energy
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mccol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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