

# Octane, 4-chloro-

<b>Other names:</b>	4-Chlorooctane
<b>Inchi:</b>	InChI=1S/C8H17Cl/c1-3-5-7-8(9)6-4-2/h8H,3-7H2,1-2H3
<b>InchiKey:</b>	SRGMFDQMGOZQPO-UHFFFAOYSA-N
<b>Formula:</b>	C8H17Cl
<b>SMILES:</b>	CCCCC(Cl)CCC
<b>Mol. weight [g/mol]:</b>	148.67
<b>CAS:</b>	999-07-5

## Physical Properties

Property code	Value	Unit	Source
gf	2.11	kJ/mol	Joback Method
hf	-229.47	kJ/mol	Joback Method
hfus	17.15	kJ/mol	Joback Method
hvap	37.40	kJ/mol	Joback Method
log10ws	-3.44		Crippen Method
logp	3.584		Crippen Method
mcvol	135.820	ml/mol	McGowan Method
pc	2462.92	kPa	Joback Method
rinpol	997.00		NIST Webbook
rinpol	997.00		NIST Webbook
ripol	1149.00		NIST Webbook
ripol	1149.00		NIST Webbook
tb	419.43	K	Joback Method
tc	594.11	K	Joback Method
tf	194.84	K	Joback Method
vc	0.526	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	262.83	J/molxK	419.43	Joback Method
cpg	275.97	J/molxK	448.54	Joback Method
cpg	288.58	J/molxK	477.66	Joback Method
cpg	300.68	J/molxK	506.77	Joback Method

cpg	312.27	J/mol×K	535.88	Joback Method
cpg	323.39	J/mol×K	564.99	Joback Method
cpg	334.03	J/mol×K	594.11	Joback Method
dvisc	0.0088497	Paxs	194.84	Joback Method
dvisc	0.0030882	Paxs	232.27	Joback Method
dvisc	0.0014434	Paxs	269.70	Joback Method
dvisc	0.0008121	Paxs	307.13	Joback Method
dvisc	0.0005177	Paxs	344.57	Joback Method
dvisc	0.0003605	Paxs	382.00	Joback Method
dvisc	0.0002677	Paxs	419.43	Joback Method

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.47709e+01
Coeff. B	-3.92812e+03
Coeff. C	-6.49020e+01
Temperature range (K), min.	336.12
Temperature range (K), max.	480.16

## Sources

**The Yaws Handbook of Vapor Pressure:**  
Crippen Method:

<https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure>  
<http://pubs.acs.org/doi/abs/10.1021/ci990307l>

**Crippen Method:**

[https://www.chemeo.com/doc/models/crippen\\_log10ws](https://www.chemeo.com/doc/models/crippen_log10ws)

**Joback Method:**

[https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)

**McGowan Method:**

<http://link.springer.com/article/10.1007/BF02311772>

**NIST Webbook:**

<http://webbook.nist.gov/cgi/cbook.cgi?ID=C999075&Units=SI>

## Legend

**cpg:** Ideal gas heat capacity  
**dvisc:** Dynamic viscosity  
**gf:** Standard Gibbs free energy of formation

<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>rinpola:</b>	Non-polar retention indices
<b>ripola:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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