

# 3'-(Trifluoromethyl)acetophenone

<b>Other names:</b>	3-Trifluoromethylacetophenone m-Trifluoromethylacetophenone 3-Trifluoromethylacetophenone Ethanone, 1-[3-(trifluoromethyl)phenyl]- 3-CF <sub>3</sub> -C <sub>6</sub> H <sub>4</sub> -COCH <sub>3</sub>
<b>Inchi:</b>	InChI=1S/C9H7F3O/c1-6(13)7-3-2-4-8(5-7)9(10,11)12/h2-5H,1H3
<b>InchiKey:</b>	ABXGMGUHGLQMAW-UHFFFAOYSA-N
<b>Formula:</b>	C <sub>9</sub> H <sub>7</sub> F <sub>3</sub> O
<b>SMILES:</b>	CC(=O)c1cccc(C(F)(F)F)c1
<b>Mol. weight [g/mol]:</b>	188.15
<b>CAS:</b>	349-76-8

## Physical Properties

Property code	Value	Unit	Source
affp	835.60	kJ/mol	NIST Webbook
basg	803.70	kJ/mol	NIST Webbook
ea	0.66 ± 0.01	eV	NIST Webbook
ea	0.77 ± 0.09	eV	NIST Webbook
gf	-582.83	kJ/mol	Joback Method
hf	-713.69	kJ/mol	Joback Method
hfus	16.14	kJ/mol	Joback Method
hvap	41.56	kJ/mol	Joback Method
log10ws	-3.24		Crippen Method
logp	2.908		Crippen Method
mcvol	120.790	ml/mol	McGowan Method
pc	3022.28	kPa	Joback Method
tb	472.00 ± 1.00	K	NIST Webbook
tc	685.09	K	Joback Method
tf	284.25	K	Joback Method
vc	0.480	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
---------------	-------	------	-----------------	--------

cpg	261.80	J/mol×K	485.43	Joback Method
cpg	273.51	J/mol×K	518.71	Joback Method
cpg	284.43	J/mol×K	551.98	Joback Method
cpg	294.60	J/mol×K	585.26	Joback Method
cpg	304.05	J/mol×K	618.54	Joback Method
cpg	312.82	J/mol×K	651.81	Joback Method
cpg	320.96	J/mol×K	685.09	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C349768&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C349768&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>affp:</b>	Proton affinity
<b>basg:</b>	Gas basicity
<b>cpg:</b>	Ideal gas heat capacity
<b>ea:</b>	Electron affinity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/46-554-6/3-Trifluoromethyl-acetophenone.pdf>

Generated by Cheméo on 2024-04-19 18:47:51.349245087 +0000 UTC m=+15841720.269822402.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.